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### (54) LABELED SUBSTRATE CONJUGATES FOR **IDENTIFYING ENZYME INHIBITORS**

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(52) **U.S. Cl.** ...... 435/7.1; 435/7.2; 435/18

Field of Classification Search ...... None See application file for complete search history.

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#### (57)**ABSTRACT**

This invention provides labeled-substrate conjugates for assaying enzymes, particularly neuraminidases. Also provided are assays that are useful for identifying compounds that inhibit sialyltransferases or neuraminidases and may be useful in treating subjects with influenza. In particular, the present invention relates to methods of using such labeled substrate conjugates to screen for enzyme inhibitors, particularly in a high-throughput format.

### 25 Claims, 20 Drawing Sheets

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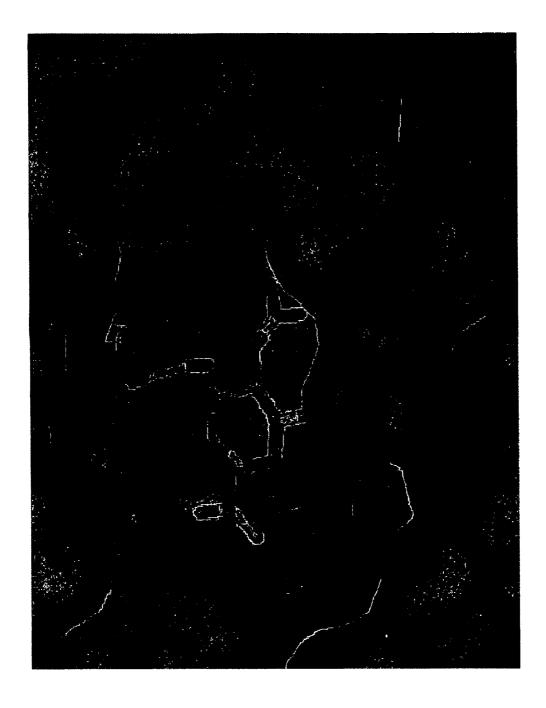
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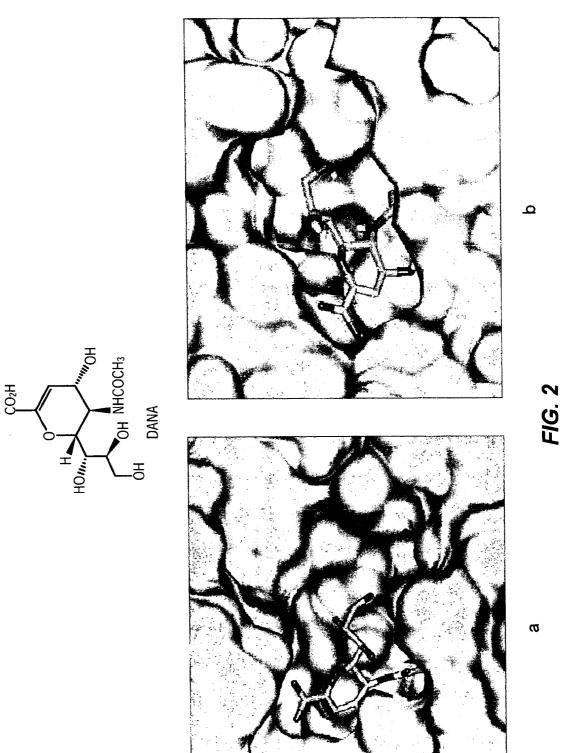
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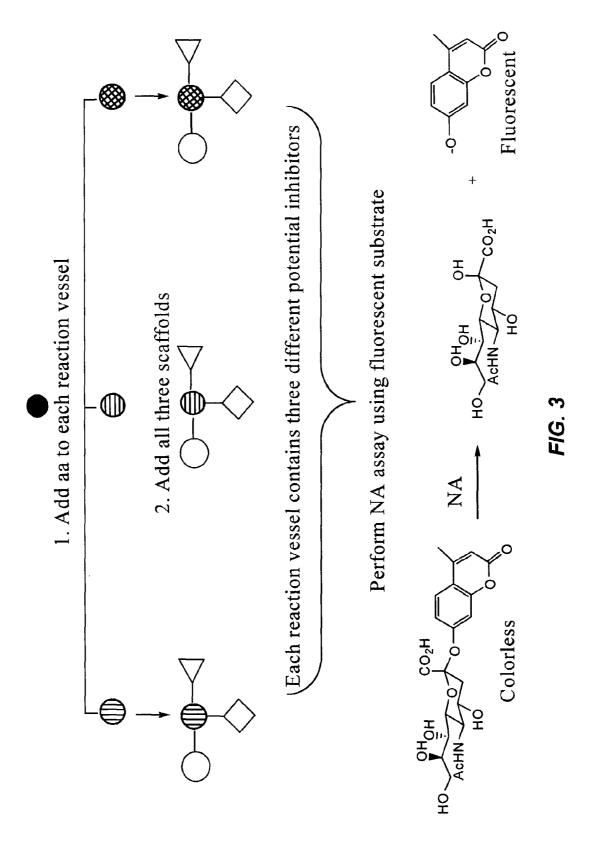
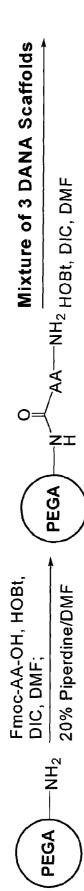
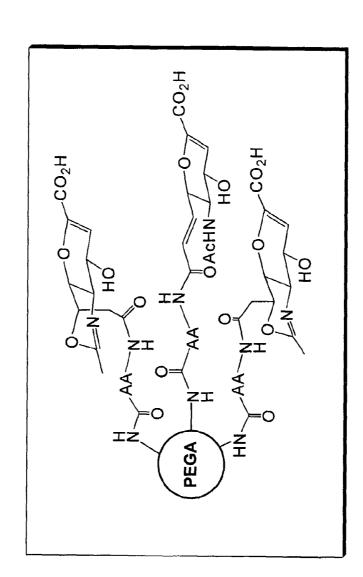
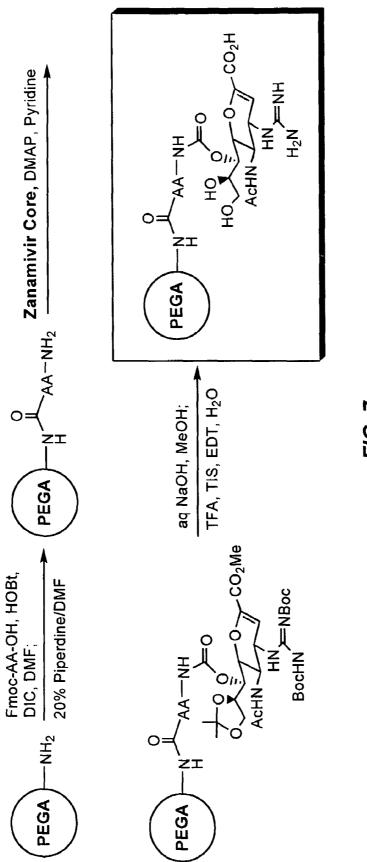


FIG. 5

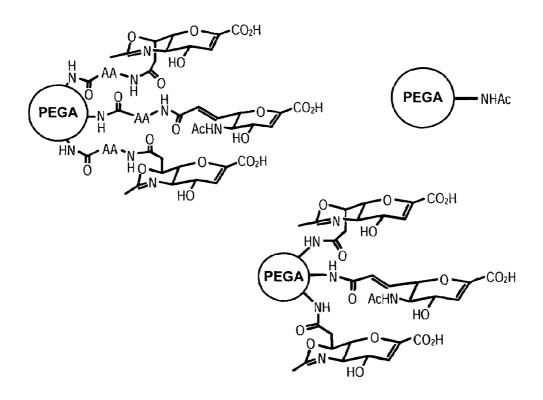




F/G. 6



F/G. 7



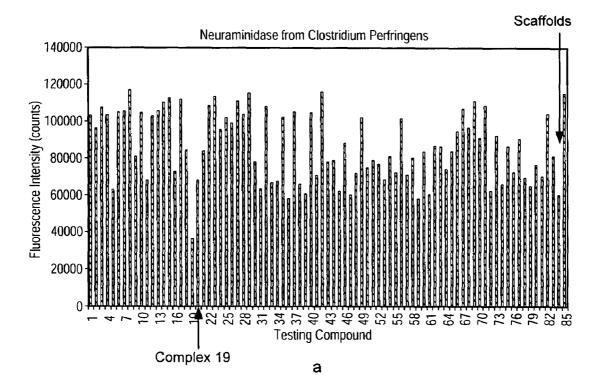
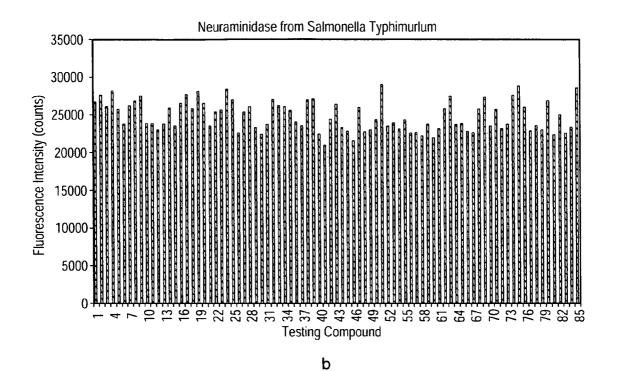


FIG. 8



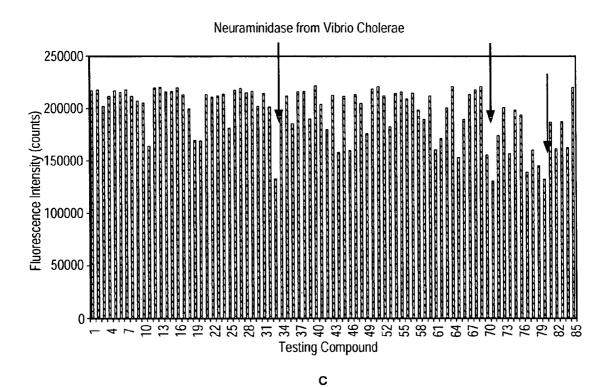
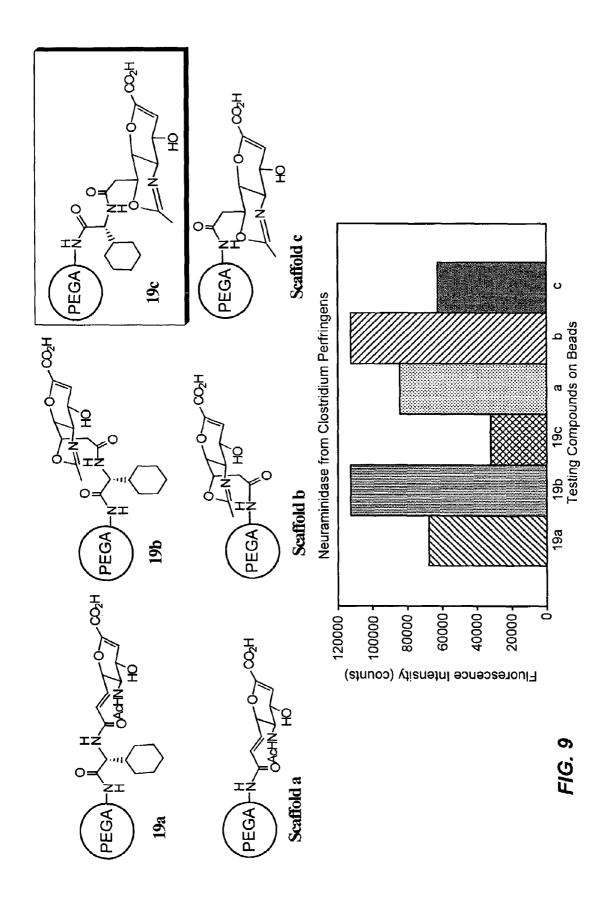
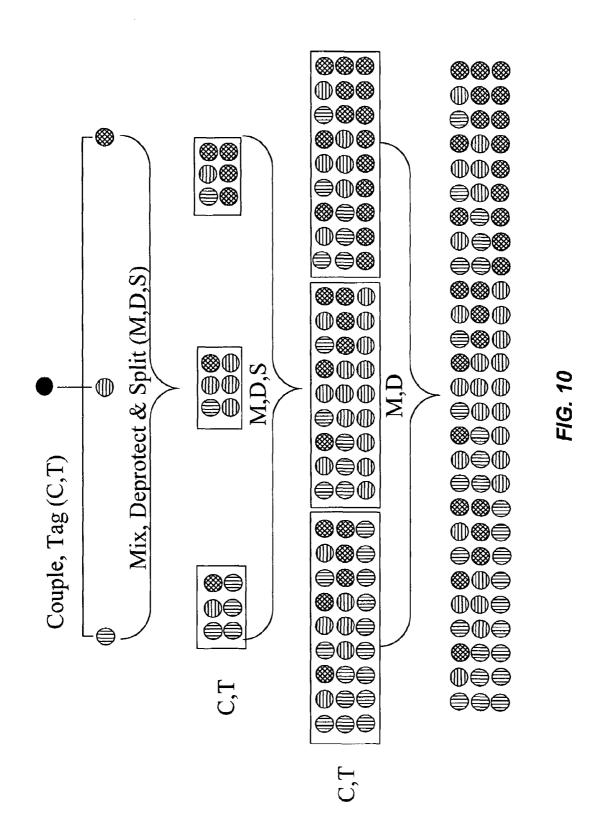


FIG. 8 (Continued)





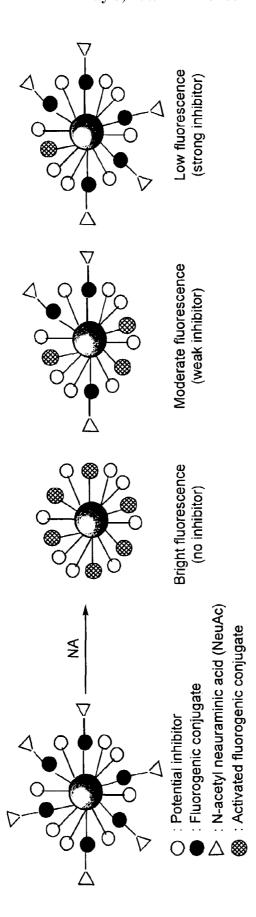
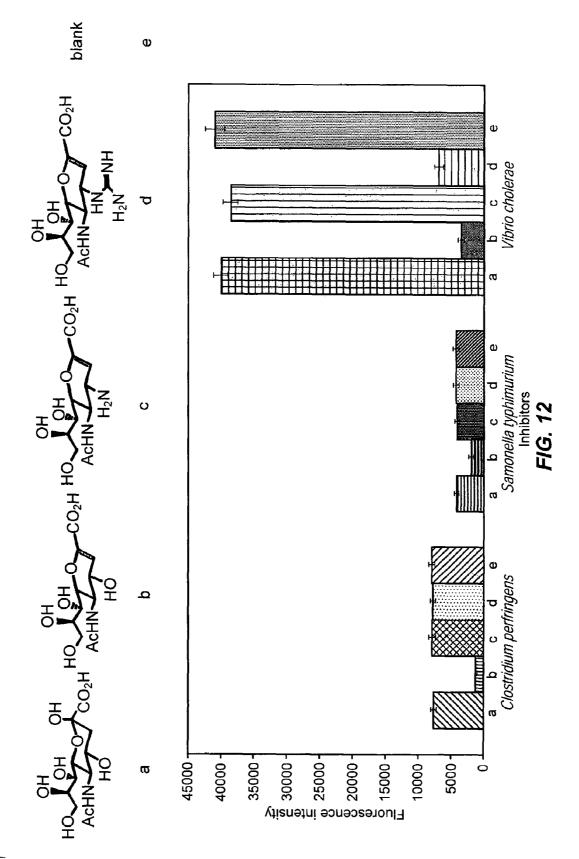
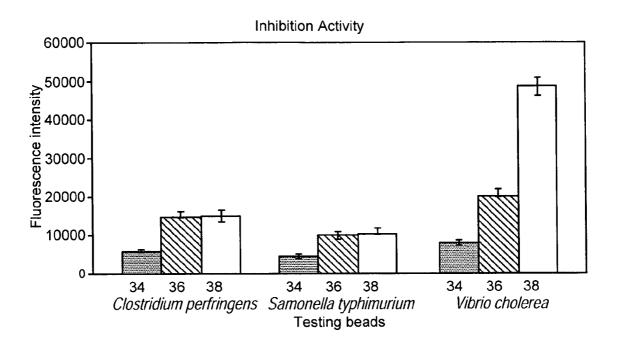


FIG. 11



4



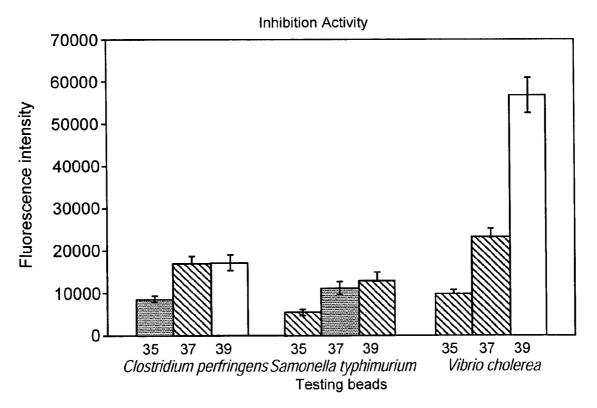
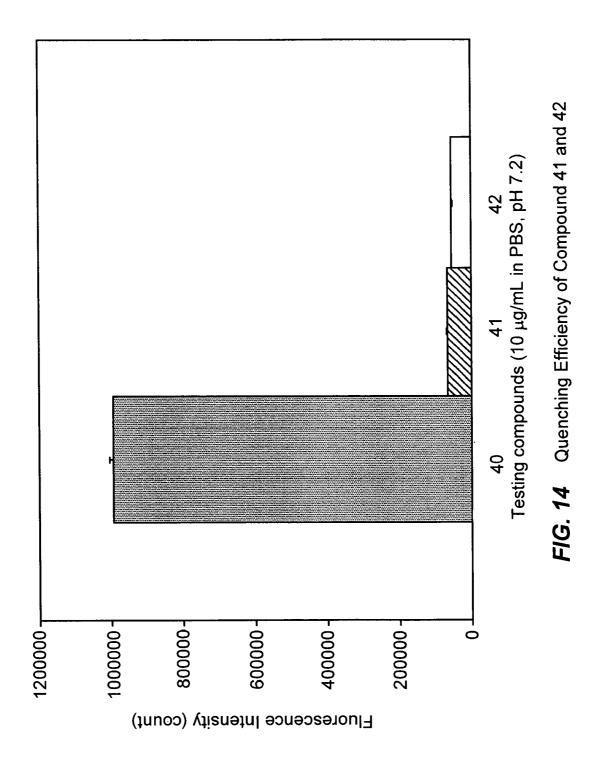
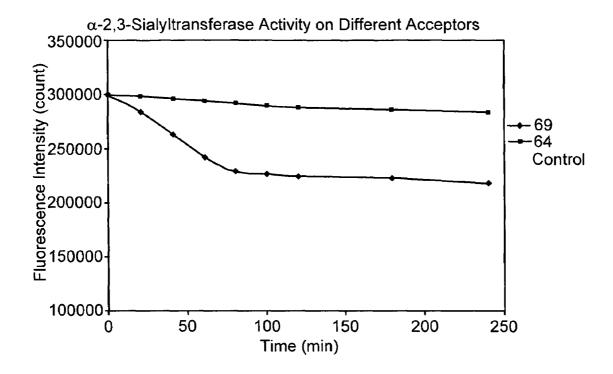


FIG. 13





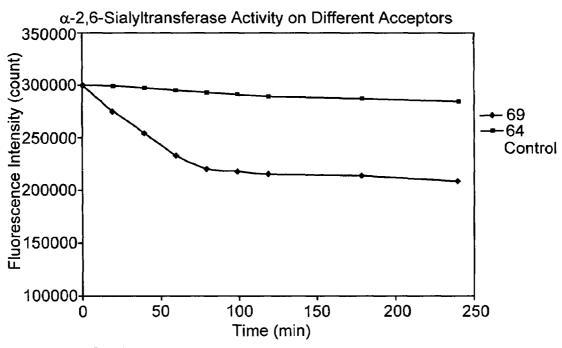


FIG. 15 Activity of Acceptors 64 and 69 for Substrate Recognition by Sialyltransferase

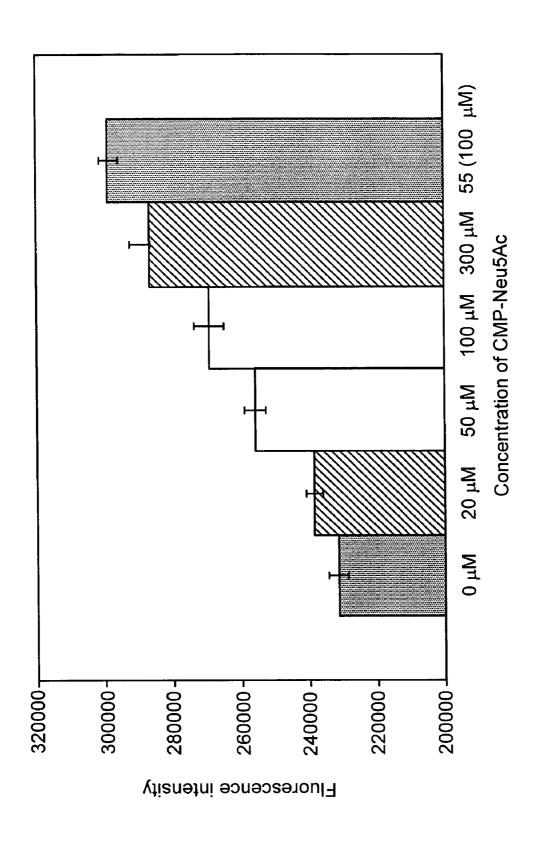


FIG. 16 Sialyltransferase Competitive Assay Uisng Donor 55 and CMP-Neu5Ac

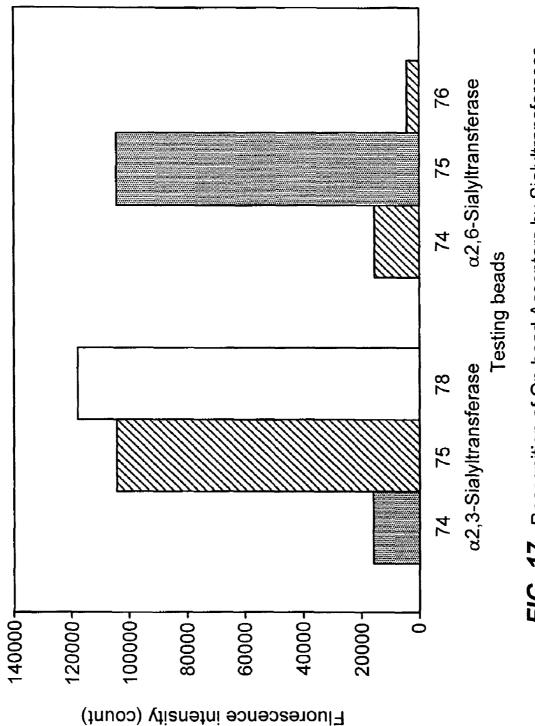


FIG. 17 Recognition of On-bead Acceptors by Sialyltransferases

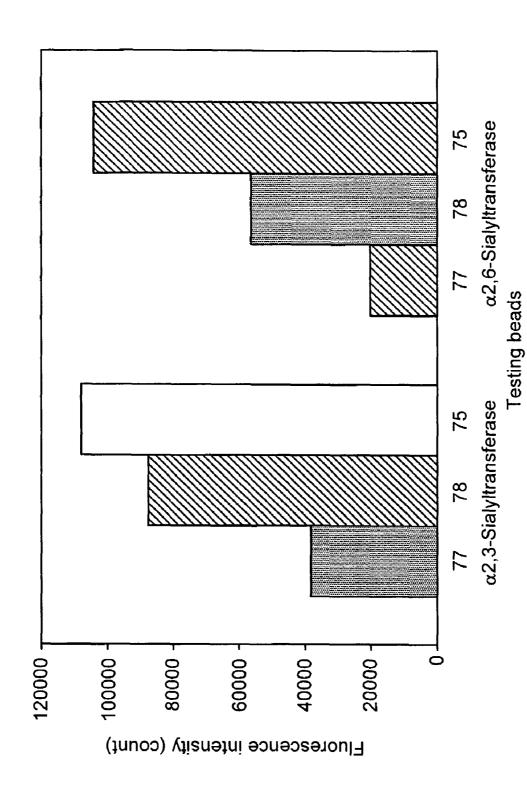
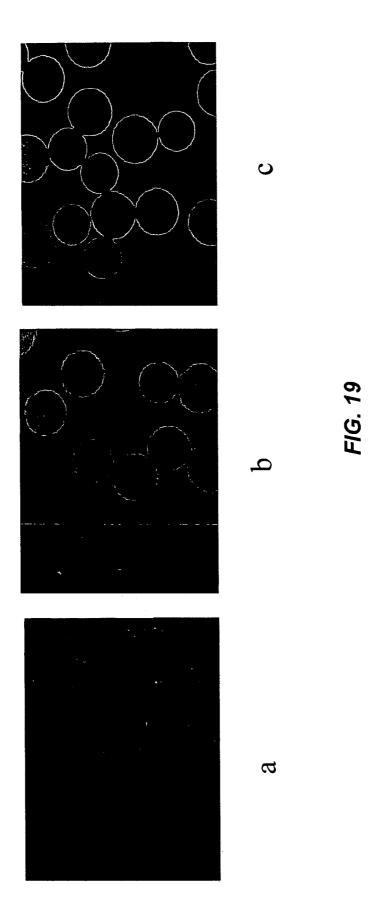


FIG. 18 Inhibitory Activity of 75, 77, and 78 Against Sialyltransferases



# LABELED SUBSTRATE CONJUGATES FOR IDENTIFYING ENZYME INHIBITORS

# CROSS-REFERENCES TO RELATED APPLICATIONS

This application claims benefit of priority under 35 U.S.C. §119(e) to U.S. Provisional application No. 60/591,414 filed Jul. 26, 2004, which is herein incorporated by reference in its entirety.

### STATEMENT AS TO RIGHTS TO INVENTIONS MADE UNDER FEDERALLY SPONSORED RESEARCH OR DEVELOPMENT

This invention was made with Government support under Grant (or Contract) No. CHE-0196482, awarded by the National Science Foundation. The Government has certain rights in this invention.

REFERENCE TO A "SEQUENCE LISTING," A TABLE, OR A COMPUTER PROGRAM LISTING APPENDIX SUBMITTED ON A COMPACT DISK

Not Applicable

### BACKGROUND OF THE INVENTION

Combinatorial chemistry has developed into a useful method for the rapid synthesis of new compounds for drug 30 discovery (see Lam, K. S. et al. *Nature* 1991, 354, 82-84). However, a crucial step in this drug discovery process is the development of high-throughput in vitro enzyme assays to identify positive hits from such combinatorial libraries.

For example, library screening has identified small molecule sulfotransferase inhibitors (see Armstrong, J. I. et al. *Angew. Chem. Int. Ed.* 2000, 39, 1303-1306). Meldal has reported using fluorescent resonance energy transfer (FRET) substrates and one-bead-two-compounds library approaches to screen protease inhibitors (see Meldal, M. *Biopolymers* 2002, 66, 93-100). A potent and specific sialyltransferase inhibitor, Soyasaponin I, has been discovered by screening over 7500 microbial extracts and natural products (see Wu, C. Y. et al. *Biochem. Biophys. Res. Commun.* 2001, 284(2), 466-469). More recently, a peptide WWWWNG-NH<sub>2</sub> was identified as a potent inhibitor for α2,3-sialyltransferase (ST3Gal I) from the screening of a combinatorial peptide library (see Lee, K. Y. et al. *J. Biol. Chem.* 2002, 277, 49341-49351).

Another pharmaceutically important enzyme is neuraminidase (NA). Neuraminidase is an enzyme that cleaves the 50 α-ketosidic linkage of the terminal sialic acids and has been found in viruses, bacteria, parasites, and mammalian cells (see Corfield, T. Glycobiology 1992, 2, 509-521; and Corfield, A. P. et al. "Role of sialidases and sialic acids in molecular recognition phenomena" in conferences Philippe Laudat 55 1991 pp. 113-134, Institute National de la Sante et de la Racherche Medicale, (INSERM), Paris; Rosenberg and Schengrund Biological roles of sialic acids Plennum Press, N.Y, 1976, pp. 295-360). It plays an important biological role in the regulation of glycoconjugates involved in cell-to-cell interactions (see Schauer, R. Adv. Carbohydr. Chem. Biochem. 1982, 40, 131-134). For example, influenza viruses have two surface glycoproteins, hemaggutinin and neuraminidase. Hemagglutinin binds to receptors containing neuraminic acid, which allows the virus to penetrate through the cell membrane. Neuraminidase destroys receptors recog- 65 nized by hemaggutinin by cleaving the  $\alpha$ -ketosidic linkage of sialic acids. This cleavage facilitates passage of the virus to

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and from sites in the respiratory tract (see Colman, P. M. "Neuraminidase: enzyme and antigen" in *The Influenza Viruses* (Klug, R. M., ed.), 1989, pp. 175-218. Plenum Publishing Corporation, New York). Studies with a neuraminidase-deficient influenza virus have shown that the mutant virus is still infective but the budding virus particles form aggregates or remain bound to the infected cell surface (see Liu, C. et al. *J. Virol.* 1995, 69, 1099-1106). Influenza causes considerable disease burden each year and while vaccination is the first line of defense against influenza A and B viruses, antiviral therapy can aid in controlling the impact of the disease (see Schmidt *Drugs* 62: 2031 (2004)).

This important biological activity has prompted chemists 15 to use the crystal structure of neuraminidase to design specific inhibitors as anti-influenza virus agents (see FIG. 1 from von Itzstein, M.; et al. Nature, 1993, 363, 418-423; and Taylhell, N. R. et al. J. Med. Chem. 1998, 41, 798-807; and FIG. 2. from Crennell, S. J. et al. J. Mol. Biol. 1996, 259, 264; and Structure <sup>20</sup> 1994, 2, 535-544 (see also Kim, C. U. et al. *J. Am. Chem. Soc.* 1997, 119, 681-690; Babu, Y. S. et al. J. Med. Chem. 2000, 43, 3482-3486; Chand, P.; et al. J. Med. Chem. 2001, 44, 4379-4392; Andrews, D. M. et al. Eur. J. Med. Chem. 1999, 34, 563-574; and Kiefel, M. J. and von Itzstein, M. Chem. Rev. 2002, 102, 471-490). Since the determination of the crystal structure of the influenza neuraminidase, (see Varghese, J. N. et al. Nature 1983, 303, 35-40; Varghese, J. N. and Colman, P. M. J. Molec. Biol. 1991, 221, 473-486; Burmeister, W. P. et al. EMBO J. 1992, 11, 49-56; and Varghese, J. N. et al. Protein Science 1995, 4, 1081-1087) many derivatives of 2-deoxy-2, 3-dehydro-N-acetylneuraminic acid (DANA) have been designed as transition state analogues to block the catalytic sites of influenza neuraminidase (see Varghese, J. N. Drug Development Res. 1999, 46, 179-196, Palese and Schulman in Chemoprophylaxis and Virus Infections of the Upper Respiratory Tract vol. 1 CRC Press Cleveland (Oxfod, J. S. ed.) 1977 pp. 189-205; and references cited therein). This has lead to the development of highly potent influenza neuraminidase inhibitors ZANAMIVIRTM and OSELTAMIVIRTM, which are currently in use for treatment of influenza virus infection (see Gubareva, L. V. et al. Lancet 2000, 355, 827-835; Holzer et al. *Glycoconj. J.* 10: 40-44 (1993); Chong et al. *Biochem.* Int. 24: 165-171 (1991); and references cited therein). More recently, the synthesis and biological evaluation of a functionalized cyclopentane analog, RWJ-270201, has been reported (Chand et al. J. Med. Chem. 44: 4379 (2001). This compound is a potent inhibitor of wild-type NA and some ZANAMIVIRTM- and OSELTAMIVIRTM-resistant influenza A and B virus variants (Gubareva, L. V. et al. Antimic. Agent and Chemother. 2001, 45, 3403-3408).

Success in achieving selective NA inhibition by modifying the glycerol side chain to increase hydrophobic interactions led to several other studies of ZANAMIVIR™ analogs. Honda et al. published a series of papers describing the syn- 40 thesis and biological evaluation of 7-O alkylated ZAN-AMIVIR™ analogs. Their first studies showed that the 7-hydroxyl could be replaced with fluorine to give an improved activity profile (see above). Methylation of the 7-hydroxyl led to slightly diminished anti-NA activity but the ethyl ether was 45 actually more active than ZANAMIVIRTM in the NA assay and both ethers showed increased activity in plaque reduction assays (see Honda et al. Biorg. Med. Chem. Lett. 12: 1921-24 (2002); Honda et al. Bioorg. Med. Chem. Lett. 12: 1925-28 (2002)). In general, compounds with alkyl ethers of less than 50 12 carbons showed potent (nM) inhibition of NA and improved activity relative to ZANAMIVIRTM in cell-culture assays. These combined studies indicate that modifications of the glycerol side-chain of ZANAMIVIRTM are tolerated and often beneficial.

Solution-phase neuraminidase inhibition assays normally use the fluorogenic substrate, 2'-(4-methylumbelliferyl)-α-D-acetyl-neuraminic acid, which is cleaved by neuraminidase to yield a fluorescent product that can be quantified using a fluorometer (Potier et al., *Anal. Biochem.* 94:287-296 (1979)), however, this assay method is not amenable to a high-throughput format. In addition, due to the fast emergence of resistant viral strains (see McKimm-Breschkin, J. L. *Antiviral Res.* 2000, 47, 1-17), there remains a need to find new influenza neuraminidase inhibitors.

Another pharmaceutically important class of enzymes are sialyl transferases. Glycoconjugates, including glycopro-

teins, glycosphingolipids, and polysaccharides, play important biological roles in cell-cell recognition, bacterial adhesion, signal transduction, and immune response (see Fukuda, M. "Roles cell surface carbohydrate" in Molecular and Cellular Glycobiology (Fukuda and Hindsgaul, eds.); Oxford University Press: New York, 2000, pp 33-44). Many of these biologically active glycans contain an essential nine-carbon sugar which is N-acetyl-neuraminic acid (NeuAc). For example, gangliosides are one of the biologically important sialylated glycosphingolipids found in vertebrate cells, nerve cells, as well as in the brain (see Vyas and Schnaar Biochimie 2000, 83, 677-682). In vivo sialylations are catalyzed by a family of enzymes known as sialyltransferases, which contribute to the diversity in the linkage and the chemical struc-<sup>15</sup> ture of sialic acid residues on cell-surface glycoconjugates. Different sialyl-linkages ( $\alpha$ -2,3,  $\alpha$ -2,6,  $\alpha$ -2,8) are elaborated by different sialyltransferases, which share the same donor substrate cytidine monophosphate-sialic acid (CMP-Neu5Ac) but differ in acceptors (see Harduin-Lepers, A. et al. Biochimie 2001, 83, 727-737). Sialyltransferase activity has been shown to correlate with cancer progression and several reasons have been postulated to explain this behavior (see Platt, F. M. et al. Science 1997, 276, 428-431). For example, sialic acids can prevent cell-cell interactions through nonspecific charge repulsion effects, which may facilitate metastasis. Secondly, sialyated glycoconjugates can be specifically bound by cell adhesion molecules such as selecting, allowing extravasation of cancer cell (see Feizi, T. *Immunol*. Rev. 2000, 173, 79-88; and Colin-Hughes, R. Biochimie 2001, 83, 667-676).

CMP-Neu5Ac

Like neuraminidases, the important biological activity of sialyltransferases has prompted chemists to design specific inhibitors of sialyltransferases for elucidating the role of sialyl residues in biological systems (see Wang, X. F. et al. *Med. Res. Rev.* 2003, 23, 32-47). Several sialyltransferase inhibitors have been developed mostly as CMP-Neu5Ac donor analogues, (see Klohs, W. D. et al. *Cancer Res.* 1979, 39, 1231-1238; Cambron, L. D. and Leskawa, K. C. *Biochem. Biophys. Res. Commun.* 1993, 93(2), 585-590; Schaub, C. et al. *Glycoconj. J.* 1998, 15(4), 345-354; Kijima-Suda, I. et al.

Cancer Res. 1986, 46, 858-862; Cohen, S B. and Halcomb, R. L. J. Org. Chem. 2000, 65, 6145-6152; Imamura, M. and Hashimoto, H. Tetrahedron Lett. 1996, 37(9), 1451-1454; Müller, B. et al. Tetrahedron Lett. 1998, 39, 509-512; Amann, F. et al. Chem. Eur. J. 1998, 4(6), 1106-1115; and Compain, P. 5 and Martin. O. V. Bioorg. Med. Chem. 2001, 9, 3077-3092), transition-state analogues, (see Müller, B. et al. Angew. Chem. Int. Ed. 1998, 37(20), 2893-2897; Schwörer, R. and Schmidt, R. R. J. Am. Chem. Soc. 2002, 124, 1632-1637; Schröder, P. N. and Giannis, A. Angew. Chem. Int. Ed. 1999, 38(10), 10 1379-1380; Sun, H. B. et al. Tetrahedron Lett. 2001, 42, 2451-2453; and Paul, P. et al. J. Biol. Chem. 1993, 268, 12933-12938, and sugar-acceptor analogues (see Kajihara, Y. et al. J. Carbohydr. Chem. 1993, 12(7), 991-996; and Carbohydr. Res. 1993, 247, 179-193).

However, all the known potent sialyltransferase inhibitors are polar and charged. These inhibitors have difficulty exerting their functions in cells or organisms due to low membrane permeability (see Platt, F. M. et al. *Science* 1997, 276, 428-431). There remains a need for the identification of cell-permeable sialyltransferase inhibitors for in vivo biological study and as pharmaceuticals.

Current sialyltransferase assays use radio-active or fluorescence labeled cytidine monophosphate N-acetylneuraminic acid (CMP-Neu5Ac) donor, and fluorescence or 25 ultraviolet (UV)-labeled acceptor (see Paulson, J. C. et al. *J. Biol. Chem.* 1977, 252, 2363-2371; Gross, H. J. et al. *Anal. Biochem.* 1990, 186, 127-134; Limberg, G. et al. *Liebigs. Ann.* 1996, 1773-1784; and Schaub, C. et al. *Glycoconj. J.* 1998, 15(4), 345-354). These assays require separation of 30 product from donor and acceptor, and are not convenient for the determination of kinetic parameters.

Combinatorial chemistry has developed into a useful method for the rapid identification of lead compounds for drug discovery (see Lam et al. Nature 1991, 354, 82-84; Lam 35 et al. Chem. Rev. 1997, 97(2), 411-448; Pirrung et al. Chem. Rev. 1997, 97(2), 473-488; Nefzi et al. Chem. Rev. 1997, 97(2), 449-472; Young et al. Curr. Opin. Drug Discovery Dev. 2004, 7(3), 318-324). The first step in the process is the facile identification of positive hits from a large collection of com- 40 pounds. In cases where libraries are prepared in solution, NA activity can be monitored using a synthetic substrate 2'-(4methylumbelliferyl)-α-D-N-acetylneuraminic acid which is cleaved to yield a fluorescent product (4-methylumbelliferone) that can be quantified fluorometrically (see Hochgurtel 45 et al. Proc. Natl. Acad. Sci. USA 2002, 99, 3382-3387; Potier et al. Anal. Biochem. 1979, 94, 287-296). Solution phase assays are not applicable to one-bead-one-compound libraries because it is impossible to identify the bead providing the activity.

Therefore there is a need for the rapid identification of new inhibitors for developing new drugs for the treatment of new and existing strains of flu virus. There also remains a need for a high-throughput screening method for enzyme inhibitors, specifically neuraminidase inhibitors and sialyltransferase 55 inhibitors. The present invention solves this problem by providing an on-bead assay of enzymes, which allows simultaneous monitoring of substrate cleavage and inhibitor efficiency. The substrates and methods for identifying enzyme inhibitors of the present invention can be used in screening of 60 large libraries of compounds for their enzyme inhibitory properties.

### BRIEF SUMMARY OF THE INVENTION

In one aspect the present invention provides a detectable label-substrate conjugate comprising a detectable label 6

covalently attached to the substrate of a enzyme. In another aspect, modification of the substrate causes a detectable change in the label.

In one aspect, the present invention provides an assay complex, such as that shown in FIG. 11, comprising a solid support having at least one test ligand attached thereto and at least one attached detectable label-substrate conjugate. The detectable label-substrate conjugate comprises a detectable label covalently attached to the substrate of an enzyme. Modification of the substrate causes a detectable change in the label.

In another aspect, the present invention provides an assay complex comprising at least one test ligand and at least one detectable label-substrate conjugate comprising a detectable label covalently attached to a first substrate of an enzyme and a quencher attached to a second substrate of said enzyme wherein modification of said substrates causes a detectable change in said label and either said detectable label covalently attached to a first substrate of said enzyme or said quencher attached to a second substrate of an enzyme is attached to a solid support.

In another aspect, the present invention relates to a method of identifying an enzyme inhibitor, specifically a neuraminidase or a sialyltransferase inhibitor, by combining in a assay mixture an assay complex of the present invention and a sufficient quantity of the enzyme to react with the substrate under noninhibitory conditions. The detection of a change in the label upon combining with the enzyme indicates the absence of an enzyme inhibitor. The assay complex is designed such that the presence of an enzyme inhibitor prevents the enzyme from modifying the substrate and causing a change in said detectable label. As such, test ligands which prevent a detectable change in the label can be identified as enzyme inhibitors.

### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 illustrates the crystal structure of ZANAMIVIR™ bound to Influenza A showing the pocket that is accessible at C-7 and, therefore, why C-7 analogs of ZANAMIVIR™ are potentially potent (von Itzstein, M. et al. *Nature*, 1993, 363, 418-423; Taylhell, N. R. et al. *J. Med. Chem.* 1998, 41, 798-807).

FIG. 2 illustrates modeled complexes of DANA with bacterial NA crystal structure showing the hydrophobic pocket that is accessible at C-7. FIG. 2a shows DANA with Samonella typhimurium Neuraminidase (Crennell, S. J. et al. J. Mol. Biol. 1996, 259, 264). FIG. 2b shows DANA with Vibrio cholerae Neuraminidase (Ca dependent) (Crennell, S. et al. Structure 1994, 2, 535-544).

FIG. 3 illustrates a parallel solid phase synthetic approach to NA inhibitors.

FIG. 4 illustrates synthetic approaches to C-7 functionalized DANA cores as bacterial neuraminidase inhibitors.

FIG. 5 illustrates synthetic approaches to C-7 functionalized ZANAMIVIR™ cores as bacterial neuraminidase inhibitors.

FIG. 6 illustrates the parallel synthesis of a library using three different DANA cores. Each bead has three different cores coupled to the same amino acid.

FIG. 7 illustrates the parallel synthesis of a library using a ZANAMIVIR<sup>TM</sup> core.

FIG. 8 illustrates the screening results of a DANA-based library. FIG. 8a identifies complex 19 and scaffolds. FIG. 8c identifies 3 possible hits with *Vibrio*.

- FIG. 9 illustrates the identification of a neuraminidase inhibitor, compound 19c.
- FIG. 10 illustrates combinatorial chemistry approaches to neuraminidase inhibitors and how it requires a method for independently screening each bead.
- FIG. 11 illustrates a high throughput on-bead screening approach for combinatorial libraries of Neuraminidase Inhibitors.
- FIG. 12 illustrates the results of a fluorescence assay of on-bead substrate 15 and 25 with inhibitors in solution. Inhibition of the cleavage of fluorogenic conjugates 15 and 25 using inhibitors a-d.
- FIG. 13 illustrates the results of an on-bead assay of bacterial neuraminidase. Fluorescence intensities of beads 35-39 incubated with three different neuraminidases.
- FIG. 14 illustrates the quenching efficiency of compound 41 and 42.
- FIG. **15** illustrates the activity of acceptors 64 and 69 for substrate recognition by sialyltransferase.
- FIG. 16 illustrates the results of a sialyltransferase competitive assay using donor 55 and CMP-Neu5Ac
- FIG. 17 illustrates the recognition of on-bead acceptors by sialyltransferases.
- FIG. 18 illustrates the inhibitory activity of 75, 77 and 78  $_{25}$  against sialyltransferases.
- FIG. **19***a* illustrates poly(ethyleneglycol) polyacrylamide (PEGA) functionalized beads 14 after incubation with *Vibrio cholerae*. FIG. **19***b* illustrates PEGA functionalized beads 15 after incubation with *Vibrio cholerae*. FIG. **19***c* illustrates PEGA functionalized beads 16 after incubation with *Vibrio cholerae*.

### DETAILED DESCRIPTION OF THE INVENTION

### Abbreviations and Definitions

The abbreviations used herein are conventional, unless otherwise defined:

Unless otherwise stated, the following terms used in the  $^{40}$  specification and claims have the meanings given below:

The term "test ligand" is used in this disclosure to describe any compound that is desirably screened for potential enzyme inhibitory activity. It includes, but is not limited to e.g. a 45 protein, oligopeptide, small organic molecule, polysaccharide, polynucleotide and the like. It can be a natural product, a synthetic compound, a chemical compound or a combination of two or more substances.

"High-throughput screening" or "HTS" refers to methods for simultaneously assaying a large number of test ligands for their ability to inhibit an enzyme, particularly neuraminidase. In general, many steps of these assays can be performed using automated equipment.

"Immobilization" refers to methods for attaching a molecule to a solid support either through covalent or non-covalent interactions with the surface of the solid support.

### General Description of the Embodiments

In one aspect, the present invention provides a detectable label-substrate conjugate comprising a detectable label which is attached to an enzyme substrate. In this aspect the detectable label is quenched when it is conjugated with the substrate. Once substrate is cleaved by the enzyme, the detectable

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label will no longer be quenched and will show a detectable change, for example, fluorescence.

In another aspect of the present invention, the detectable label-substrate conjugate is used in solution in combination with a test ligand for high-throughput screening of potential enzyme inhibitors. The test ligands may be in solution or immobilized on solid phase supports (see e.g. FIG. 3).

In another aspect of the present invention, the detectable label-substrate conjugate is attached to solid support in combination with a test ligand for high-throughput screening of potential enzyme inhibitors.

In another aspect, the present invention provides a method for identifying potential enzyme inhibitors, specifically neuraminidase or sialyltransferase inhibitors (see e.g. FIG. 11). In this method, a detectable label-substrate conjugate and a test ligand are combined with an enzyme which may modify the substrate. This modification may take the form of cleaving the substrate from the detectable label or ligating a second substrate onto the first substrate attached to the detectable label. If the test ligand shows no inhibition, the enzyme will quickly cleave the substrate from the detectable label-substrate complex or ligate a second substrate onto the detectable label-substrate complex. Either scenario causes a detectable change in the detectable label, such as fluorescence. If the test ligand is a strong inhibitor of the enzyme, the detectable label does not change and the support containing the test ligand will be picked out as positive hit for structure characterization. The test ligands may be in solution or immobilized on solid phase supports (see e.g. FIG. 3).

Another aspect of the present invention is to provide efficient HTS methods for screening enzyme inhibitors both in 35 solution and on solid-support.

# Detectable Label-Substrate Conjugates and their Synthesis

In one aspect, the present invention provides a detectable label-substrate conjugate comprising a detectable label which is attached to an enzyme substrate. In this aspect the detectable label is quenched when it is conjugated with the substrate. Once substrate is cleaved by the enzyme, the detectable label will no longer be quenched and will show a detectable change, for example, fluorescence.

For the purposes of the present invention suitable detectable labels, include but are not limited to fluorophores, colorometric labels, radiolabels, and the like.

For the purposes of the present invention any enzyme substrate may be used. Preferred substrates depend on the particular application of the detectable label-substrate conjugate. For example, if the detectable label-substrate complex is to be used in an enzyme assay, preferred substrates are the natural substrates of the enzyme to be assayed. For example if the enzyme to be assayed is a neuraminidase, then examples of suitable substrates include, but are not limited to sialic acid, and the like. If the enzyme to be assayed is a sialyltransferase then examples of suitable substrates include, but are not limited to sialic acid and the like. Other enzyme and substrate combinations are well known to those of skill in the art.

For ease of illustration, this aspect of the present invention will first be illustrated by conjugates for use in a neuraminidase assay. The synthesis of an on-bead fluorogenic neuraminidase substrate conjugate is shown in Scheme 1. 7-Hydroxy-4-methylcoumarin 1 is

HO 
$$\frac{a}{96\%}$$
  $\frac{b}{88\%}$   $\frac{c}{78\%}$ 

AcO 
$$\xrightarrow{\text{O}}$$
 O  $\xrightarrow{\text{HO}}$  O  $\xrightarrow{\text{E}}$  O  $\xrightarrow{\text{Br}}$  O  $\xrightarrow{\text{E}}$  O

HO AcHN HO 
$$OH$$
 OH  $OH$  OH  $O$ 

- a. Allyl bromide,  $K_2CO_3$ , acetone;
- b. N,N-Diethylaniline;
- c. 2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-N-acetylneuraminic acid, Ag<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>CN;
- d. RuO<sub>2</sub>, NaIO<sub>4</sub>,CCl<sub>4</sub>/CH<sub>3</sub>CN/H<sub>2</sub>O;
- e. NaOMe, MeOH;
- f. DIEA, DMF;
- g. NaOMe, MeOH; aq. NaOH;
- h. 1,3-Acetonedicarboxylic acid, 70% H<sub>2</sub>SO<sub>4</sub>; SOCl<sub>2</sub>, MeOH;
- i. HOBt, DIC, DMF.

reacted with allyl bromide in the presence of K<sub>2</sub>CO<sub>3</sub> to give 2 (see Clarke, D. J. and Robinson, R. S. Tetrahedron 2002, 58, 2831-2837) which can be subsequently rearranged under microwave conditions to afford regioselective product 3 (see Saidi, M. R. and Rajabi, F. Heterocycles 2001, 55, 1805-1812) Glycosylation of 2-deoxy-2-chloro-4,7,8,9-tetra-Oacetyl-N-acetylneuraminic acid methyl ester with 3 using silver carbonate in the presence of activated molecular sieves gives 4 (see Potier, M. et al. Anal. Biochem. 1979, 94, 287-296). The acetyl group of substrate 4-bromomethyl-7-acetoxycoumarin 6 is quantitatively removed with sodium methoxide, followed by glycosylation with 2-deoxy-2chloro-4,7,8,9-tetra-O-acetyl-N-acetylneuraminic methyl ester to give the protected substrate 8. The substrate 8 can be immobilized on a resin before removal of the protect- 65 ing group. Unprotected substrate 14 can also be synthesized from resorcinol 11. Resocinol 11 is first converted to 7-hy-

droxycoumarin-4-acetic acid methyl ester 12 (see Zhu, Q. et al. *Organic Lett.* 2003, 5, 1257-1260). Glycosylation of 2-deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-N-acetyl-

neuraminic acid methyl ester with 12 gave protected substrate 13, followed by deprotection to yield 14.

Substrates can be immobilized on solid supports by methods known in the art. For example, 14 can be immobilized on a resin by standard amide coupling to give 15 (see e.g. M. Meldal *Tet. Lett.* 33: 3077 (1992)).

For the purposes of the present invention suitable solid supports include, but are not limited to beads, slides, chips, and the like.

Confirmation of immobilization can be accomplished by treatment with the appropriate enzyme. For example, the resin 15 showed strong blue fluorescence once it was incubated with neuraminidase.

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Other labels can also be used in the detectable label-substrate conjugates of the present invention. For example, as shown in Scheme 2, Fluorescein labeled substrates can also be used as a fluorogenic substrate. Fluorescein 16 is first converted to 17, followed by deprotection of allyl ester to yield 18 (see Zaikova, T. O. et al. *Bioconjugate Chem.* 2001, 12, 307-313). Glycosylation of 18 with 2-deoxy-2-chloro-4, 7,8,9-tetra-O-acetyl-N-acetylneuraminic acid methyl ester affords protected substrate 19. Fluorescein 16 can be selectively protected to give the allyl ester 21, followed by alkyla-

tion with methyl bromoacetate and deprotection of allyl ester to yield 22 (see Grieco, P. et al. *J. Peptide Research* 2001, 57(3), 250-256). Glycosylation of 22 with 2-deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-N-acetylneuraminic acid methyl ester gives protected substrate 23, which can be deprotected to afford the substrate 24. The substrate 24 was immobilized on the resin by standard amide coupling to give 25. The beads showed strong green fluorescence once the beads were incubated with neuraminidase.

#### -continued

a. Allyl bromide, K2CO3, DMF;

b. aq. NaOH/MeOH;

c. 2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-N-acetylneuraminic acid, Ag<sub>2</sub>CO<sub>3</sub>, CH<sub>3</sub>CN;

d. RuO2, NaIO4, CCl4/CH3CN/H2O;

e. Methyl bromoacetate, K2CO3, DMF; Pd(PPh3)4, PhSiH3, CH2Cl2

f. NaOMe, MeOH; aq. NaOH;

g. HOBt, DIC, DMF

### Synthesis of Test Ligands

The ligands to be tested using the methods of this invention as possible enzyme inhibitors are available from a wide variety of sources or may be synthesized by a wide variety of 50 methods known by those skilled in the art. These sources include, but are not limited to, commercially available chemical clearing houses, Sigma, Aldrich, pharmaceutical companies. The composition of these compounds can be wide ranging. They can be organic small molecules, inorganic molecules, carbohydrates, peptides, nucleic acids and mixtures of compounds. The test ligands can be single compounds or multiple compounds such as libraries. Test ligands may be made by traditional synthesis or by combinatorial chemical techniques known in the art of organic synthesis (see FIG. 10).

For example, Schemes 3-4 shows the synthesis of an analog 29 of known neuraminidase inhibitor 2-deoxy-2,3-dehydro-N-acetylneuraminic acid (FIG. 12b) and analog 33 of known neuraminidase inhibitor 4-guanidino-2-deoxy-2,3-didehydro-N-acetylneuraminic acid (Neu5Ac2en) (FIG. 12d), and their attachment to a bead solid support together with a detectable label-substrate conjugate. As shown in Scheme 3,

the 8,9-dihydroxy groups of 2-deoxy-2,3-dehydro-N-acetylneuraminic acid methyl ester 26 (see Kirchner, E. et al. J. Carbohydr. Chem. 1988, 7, 453-86) can be selectively protected with an isopropylidene protecting group, followed by protection of the 4-hydroxy group with tert-butyldimethylsilyl chloride (TBSCl) to give the intermediate 27 (see Wyatt, P. G. et al. Bioorg. Med. Chem. Lett. 2001, 11, 669-673). The 7-hydroxy group of 27 can then be alkylated with methyl bromoacetate to afford 28 (see Masuda, T. et al. Bioorg. Med. Chem. Lett. 2003, 13, 669-673). Both of the methyl esters can be removed by aqueous sodium hydroxyide (NaOH) to yield

Scheme 3.

26

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a. 2,2-Dimethoxy propane, acetone, p-TsOH; TBSCI, imidazole, DMF; b. Methyl bromoacetate, NaH, DMF; c. aq NaOH, THF.

The synthesis of 4-guanidino-Neu5Ac2en (FIG. 12d) derivative 33 is shown in Scheme 4. The azide 30 can be reduced by hydrogenation over Lindar catalyst, followed by protection of the 4-amino group by guanidinylation with 35 N,N'-bis-tert-butoxycarbonyl-1H-pyrazole-1-carboxamidine (BisBocPCH) to produce the protected guanidine 31 (see Bernatowicz, M. S. et al. Tetrahedron Lett. 1993, 34, 3389-3392). After removal of the acetate protecting group using catalytic sodium methoxide in methanol, the 8,9-dihydroxy groups were selectively protected using 2,2-dimethoxy propane and catalytic p-toluenesulfonic acid in acetone to give the 8,9-isopropylidene protected intermediate 32 (see Kirchner, E. et al. J. Carbohydr. Chem. 1988, 7, 453-86). Compound 32 can then be treated with 4-nitrophenyl chloroformate and DMAP in dry pyridine to generate the active ester  $33^{-45}$ (see Andrews, D. M. et al. Eur. J. Med. Chem. 1999, 34, 563-574).

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-continued

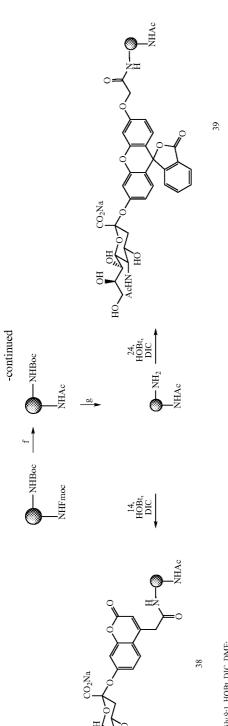
a. Lindar cat., H2, EtOH; N,N'-bis(tert-butoxycarbonyl)-1H-pyrazole-1-carboxamidine, TEA, THF;

b. NaOMe, MeOH; 2,2-dimethoxy propane, acetone, p-TsOH; c. p-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>OCOCl, DMAP, Pyridine.

Similarly, FIGS. 4 and 5 illustrate synthetic approaches to C-7 functionalized DANA and ZANAMIVIR<sup>TM</sup> cores, respectively, as bacterial neuraminidase inhibitors.

The test ligands can be used in solution or attached to a solid support and Scheme 5 illustrates one embodiment synthesizing one-inhibitor/one-substrate functionalized beads. For example, PEGA resin can be functionalized with orthogonal protecting groups by coupling with a mixture of 9-fluorenylmethoxycarbonyl (Fmoc) protected glycine and tert-butyloxycarbonyl (Boc) protected glycine (9:1) (Scheme 5) (see also Liu et al. J. Am. Chem. Soc., 2002, 124, 7678-7680; Bennett et al.; Saneii, H. H. eds. In Advanced ChemTech Handbook of combinatorial and solid phase organic chemistry—a guide to principles, products and pro-50 tocols. 1998, pp 330, Advanced Chemtech). After removal of Fmoc protecting group using 20% piperidine in dimethylformamide (DMF), compound 29 can be coupled to the beads using 1-hydroxybenzotriazole (HOBt) and diisopropylcarbodiimide (DIC). The tertiary-butyl silyl (TBS) group can be deprotected using tetrabutyl ammonium fluoride (TBAF) in tetrahydrofuran (THF). After removal of tertiary-butoxy carbonyl (Boc) and isopropylidene groups using trifluoroacetic acid (TFA) in dichloromethane (DCM), the beads can be coupled with the fluorogenic substrates 14 and 24 individually. Using the same methodology, compound 33 can also immobilized on orthogonally protected resin. The resin can then be conjugated with substrates 14 and 24 individually. Orthogonally protected resin that is also conjugated with

substrates 14 and 24 after acylation can be used as a control.



a. Fmoc-Gly/Bac-Gly 9:1, HOBŁ DIC, DMF;
b. 20% piperdine in DMF; 29, HOBŁ, DIC, DMF;
c. TBAF, THF, 50%, TFA in DCM;
d. 20% piperdine in DMF; 33, DMAP, pyridine;
e. aq. NoPH, MOCH; 30% TFA in DCM;
f. 20% nineridine in DMF Ac.O. worldine.

Similarly, FIGS. **6** and **7** illustrate parallel synthetic approaches to libraries using functionalized DANA and ZANAMIVIR<sup>TM</sup> cores, respectively. Each support can be coupled to a plurality of different cores.

The components in solution or the functionalized resins 5 can then be incubated with enzyme in a suitable buffer solution. The fluorescence intensity was measured using a fluorescence plate reader after a sufficient incubation period. One-bead-one-ligand-one-substrate combinatorial libraries may be used in a high-throughput screening method for identifying the test ligands as enzyme inhibitors. In another aspect of the invention, one-bead-multiple ligand-multiple substrate combinatorial libraries may also be made and used in a HTS method as outline above.

These complexes and methods are also suitable for 15 enzymes that have multiple substrates. Thus in another aspect of the present invention, the detectable label may be quenched when it is brought into proximity of a quencher conjugated with the substrate. If the substrate is cleaved by the enzyme, the quencher will diffuse away from the detectable label and 20 the detectable label will show a detectable change, for example, fluorescence. If the substrate is ligated by the enzyme, the quencher will be brought in proximity of the detectable label and the detectable label will show a detectable change, for example, quenching of fluorescence.

To illustrate this aspect of the present invention, the enzyme sialyltransferase will be used, however, the enzyme system used is not limiting to the present invention. For example as shown in Scheme 6, a "donor" substrate, e.g. CMP-Neu5Ac, which is conjugated with a detectable label 30 (e.g. a fluorescent label), and a "acceptor" substrate is modified with a quencher (e.g. a fluorescent quencher). When the

donor and acceptor substrates are incubated with their enzyme, the product will become non-detectable, e.g. non-fluorescent. The detectable change in the label, e.g. fluorescence to non-fluorescence, can be used to monitor the activity of the enzyme, e.g. sialyltransferase.

Examples of detectable label-quencher molecules include FRET pairs, and the like. Examples of FRET pairs include, but are not limited to, o-aminobenzamide (Abz)-3-nitrotyrosine [Tyr(NO<sub>2</sub>)], (see Meldal, M. et al. *Proc. Natl. Acad. Sci. USA.* 1994, 91, 3314-3318); Abz-ethylenediamine dinitrophenyl (EDDnp) (see Chagas, J. R. et al. *Anal. Biochem.* 1991, 192, 419-425; and Juliano L. et al. *Biochem. Biophys. Res. Commun.* 1990, 173, 647-652); and 5-[(2'-aminoethyl) amino]naphthalene sulphonic acid (EDANS)-4-[[4'-(dimethylamino)phenyl]azo]benzoic acid (DABCYL) (see Taliani, M. et al. *Lett. Pept. Sci.* 1997, 4, 101-106).

In one aspect of the present invention, the testing ligand is first incubated with enzyme, e.g. sialyltransferase, and then the donor and acceptor substrates are added to the mixture. In this aspect, the inhibitory activity can be analyzed based on the detectable change in the label, e.g. fluorescence.

In another aspect of the invention, one-bead-one-compound-one-acceptor libraries can be constructed. The library is then incubated with enzyme, e.g. sialyltransferase, in the presence of a donor substrate. If the test ligand on the bead does not inhibit the enzyme, the donor substrate will be transferred to acceptor substrate, and the bead will show no change in the detectable label, e.g. strong fluorescence. If the test ligand on bead is a potent inhibitor, it will block the transfer of the donor substrate to the acceptor substrate, and the detectable label will remain dark.

Scheme 6. HTS of Sialyltransferase Inhibitors Both in Solution and on Soild-support

## -continued

Fluorescent bead

In another aspect of the invention, one-bead-one-compound-one-donor libraries can be constructed. The library is then incubated with enzyme, e.g. sialyltransferase, in the presence of a acceptor substrate. If the test ligand on the bead does not inhibit the enzyme, the donor substrate will be transferred to acceptor substrate, and the bead will show no change in the detectable label, e.g. strong fluorescence. If the test ligand on bead is a potent inhibitor, it will block the transfer of the donor substrate to the acceptor substrate, and the detectable label will remain dark.

If FRET pairs are used as the detectable label, it is recognized that suitable FRET pairs may depend on the size and the tendency of the FRET pairs to interact with the particular enzyme, and the efficiency of energy transfer. For sialyltransferase, 7-hydroxycoumarin-4-acetic acid 40 conjugated with Tyr(NO<sub>2</sub>) exhibits sufficient quenching of fluorescence. Likewise, as shown in FIG. **14**, compounds 41 and 42 showed excellent fluorescence quench, the quenching efficiency being over 90%.

40 
$$\frac{1}{1}$$
  $\frac{1}{1}$   $\frac$ 

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\$$

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## Synthesis of Detectable Label-Donor Substrate Conjugates

The synthesis of suitable detectable label-donor substrate conjugates for sialyltransferase is exemplified below in Scheme 7. To make the fluorescent donor, 2-deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-N-acetylneuraminic acid 43 (see Marra, A. and Sinay, P. Carbohydr. Res. 1989, 190, 317-322) 10 can be treated with 4-methoxybenzyl alcohol (PMBOH) in the presence of silver carbonate to give 49 (see Ikeda, K. et al. Bioorg. Med. Chem. Lett. 2002, 12, 2309-2311). The replacement of the N-acetyl group of 49 with a fluorescence label can give 50. Deprotection of 4-(para)-methoxybenzyl (PMB) group of 50 by 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) yields 51 (see Horita, K. et al. Tetrahedron 1986, 42, 3021-3028). Reaction of 51 with chlorodiethoxyphosphane in the presence of Hünig's base affords the β-phosphite, fol- 20 lowed by reaction with N,O-acetyl-protected CMP derivative 52 (see Martin, T. J. et al. Bioorg. Med. Chem. 1994, 2, 1203-1208). gives the corresponding CMP-Neu5Ac analogue.

Scheme 7.
Synthesis of Fluorescent CMP-Neu5Ac Analogues

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An alternative synthetic approach is illustrated in Scheme 8. The N-acetyl group of 49 can first be replaced by 9-fluorenylmethoxycarbonyl (Fmoc)-protected glycine to afford 53. After removal of the Fmoc protecting group, the immediate can be conjugated with compound 45 to give 54. Deprotection of PMB group of 54 by DDQ, and reaction of anomeric hydroxy group with chlorodiethoxyphosphane in the presence of Hünig's base, then conjugation of the  $\beta$ -phosphite with N,O-acetyl-protected CMP derivative 52 (see Martin, T. J. et al. *Bioorg. Med. Chem.* 1994, 2, 1203-1208), gives the corresponding CMP-Neu5Ac analogue. De-O-acetylation with sodium methoxide (NaOMe) in methanol, followed by removal of N-acetyl protecting group and methyl ester with NaOH in water yields the fluorescent CMP-Neu5Ac analogue 55.

-continued

More diverse fluorescent CMP-Neu5Ac analogues can be 20 made as shown in Scheme 9. The N-acetyl group of 44 can be removed by phosphorus pentachloride, (see Ikeda, K. et al. Chem. Pharm. Bull. 1991, 39, 1305-1309), followed by acylation with trifluoroacetylamino-acetyl chloride (see Nordlander, J. E. et al. J. Org. Chem. 1984, 49, 4107-4111), to give 56. Hydrogenolysis of 56 in the presence of Pd/C, and reaction of anomeric hydroxy group with chlorodiethoxyphosphane in the presence of Hünig's base, then conjugation of the β-phosphite with N,O-acetyl-protected CMP derivative 52 (see Martin, T. J. et al. Bioorg. Med. Chem. 1994, 2, 1203-1208), affords the corresponding CMP-Neu5Ac analogue 57 (see Dufner, G. et al. Eur. J. Org. Chem. 2000, 1467-1482) Removal of all protecting groups under the condition described above gives 58. Conjugation of 58 with fluoresceinyl isothiocyanate (FITC) in DMF and water at pH 8~9 affords 59 (see Brossmer, R. and Gross, H. J. Methods Enzymol. 1994, 247 B, 177-192). Reaction of 58 with 7-hydroxy-4-coumarinylacetic acid N-succinimidyl ester (see Demant, E. J. F. Biochimica et Biophysica Acta 1996, 1304, 43-55), in dimethylsulfoxide (DMSO) and water at pH 8~9 affords 55. Treatment of 58 with 7-amino-4-methyl-3-coumarinacetic acid N-succinimidyl ester (see Stefanova, H. I. et al. Biochem. 1993, 32, 356-62), in DMSO and water at pH 8~9 affords 55.

## Scheme 9. Synthesis of Diverse Fluorescent CMP-Neu5Ac Analogues

## -continued

$$\begin{array}{c} & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

$$\begin{array}{c} \text{NH}_2 \\ \text{OH} \\ \text{OO} \\ \text{N} \\ \text{OO} \\ \text{ONa} \\ \text{OH} \\ \text{OH}$$

## Synthesis of Quencher-Acceptor Substrate Conjugates

Examples of the synthesis the suitable quencher-acceptor substrate conjugates for sialyltransferases is shown in Scheme 10. Glycosylation of lactose peracetate 61 (SIGMA) with 2-azidoethanol (see Chernyak, A. Y. et al. *Carbohydr.* 65 *Res.* 1992, 223, 303-309) in the presence of boron trifluoride etherate gives the  $\beta$ -glycoside 62 (see Sun, X. L. et al. *Biom*-

acromolecules 2002, 3, 1065-1070). Hydrogenation of 62 in the presence of palladium on carbon (Pd/C) affords 63 (see Hatanaka, Y. et al. *Bioorg. Med. Chem. Lett.* 1995, 5, 2859-2862). Reaction of 63 with  $N^{\alpha}$ -Fmoc-3-nitro-L-tyrosine pentafluorophenyl ester [Fmoc-Tyr(NO<sub>2</sub>)-OPfP], followed by removal of the Fmoc protecting group with 20% piperidine in DMF and the acetate protecting groups with NaOMe in methanol (MeOH) gives 64.

$$\begin{array}{c} OAc \\ OAc \\ OAc \\ OAc \\ OAc \\ OAc \\ \end{array} \\ \begin{array}{c} OAc \\ OAc \\ OAc \\ \end{array} \\ \begin{array}{c} OAc \\ OAc \\ OAc \\ \end{array} \\ \begin{array}{c} Pd/C, H_2, MeOH \\ 82\% \\ \end{array} \\ \end{array}$$

N-Acetyllactosamine is an acceptor substrate for both  $\alpha$ 2,3- and  $\alpha$ 2,6-sialyltransferases (see Limberg, G. et al. *Lie*bigs Ann. 1996, 1773-1784) and the synthesis of N-acetyllactosamine derived acceptor is shown in Scheme 11. Treatment of N-acetyllactosamine peracetate 65 (SIGMA) with acetyl chloride and small amount of concentrated hydrogen chloride (HCl) gives halide 66 (see Kaifu, R. and Osawa, T. Carbohydr. Res. 1976, 52, 179-185). Reaction of halide 66 with <sup>20</sup> 2-azidoethanol (see Chernyak, A. Y. et al. *Carbohydr. Res.* 1992, 223, 303-309) in the presence of mercuric cyanide affords 67 (see Mohan, H. et al. Synlett 2003, 9, 1255-1256; Rana, S. S. and Matta, K. L. Carbohydr. Res. 1983, 113, C18-C21). Treatment of 67 with catalytic NaOMe in MeOH gives 68 (see Blixt, O. et al. J. Org. Chem. 2001, 66, 2442-2448). Reduction of 67 with hydrogen in the presence of Pd/C, then conjugation with Fmoc-Tyr(NO<sub>2</sub>)-OPfp (pentafluorophenyl ester), followed by removal of the Fmoc protecting group with 20% piperidine in DMF and the acetate protecting groups with NaOMe in MeOH gives 69.

#### Scheme 11 Synthesis of a N-Acetyllactosamine-Derived Acceptor

The synthesis of 2-acetamido-2-deoxy-3-O-β-D-galacto-pyranosyl-β-D-glucose derived acceptor is shown in Scheme 12. Treatment of 70 (ACROS) with benzaldehyde dimethylacetal in the presence of a catalytic amount of p-toluenesulfonic acid gives 71 (see Jeanloz, R. W. et al. *Carbohydr. 5 Res.* 1968, 6, 184-196). Glycosylation of 71 with 2,3,4,6-tetra-O-acetyl-α-D-galactopyranosyl bromide in 1:1 nitrobenzene/benzene in the presence of mercuric cyanide affords 72 (see Matta, K. L. and Barlow, J. J. *Carbohydr. Res.* 1975, 43, 299-304). Hydrogenolysis of 72 in the presence of 10 Pd/C gives 73.

Soc. 2000, 122, 11270-11271), which does not interfere with fluorescence of donor. After removal of Fmoc protecting group using 20% piperidine in DMF, the resin was then conjugated with a mixture of  $N^{\alpha}$ -Fmoc/acyl (Ac)-glycine (1:10) to decrease the loading. After deprotection of Fmoc protecting group again, the resin can be reacted with succinic anhydride. The carboxylic acid on bead can then be conjugated with acceptor 63, followed by removal of acetate group using NaOMe in MeOH to give 74. To immobilize the acceptor 68 on bead, the resin can be coupled with a mixture of 4-pentynoic acid/ $N^{\alpha}$ -Ac-glycine (1:10). The terminal alkyne on the

Scheme 12. Synthesis of 2-Acetamido-2-deoxy-3-O-β-D-galactopyranosyl-β-D-glucose Derived Acceptor

Either the acceptor or the donor may be immobilized on the solid support. The solid support used is not limiting. For example, acryloylated bis(2-aminopropyl)polyethylene gly-col/dimethyl acrylamide copolymer (PL-PEGA) resin can be used as solid support, which can be swelled in both organic and aqueous media. The PEGA resin was first coupled with 65 Fmoc-4-NO<sub>2</sub>-phenylalanine to reduce the auto-fluorescence of the beads (Scheme 13), (see Harris, R. F. et al. *J. Am. Chem.* 

bead can be subjected to 1,3-dipolar cycloaddition with acceptor 68 in the presence of copper sulfate (CuSO<sub>4</sub>) and sodium ascorbate to give 75 (see Rostovsev, V. V. et al. *Angew. Chem. Int. Ed.* 2002, 41, 2596-2599). To synthesize the acceptor 76, the carboxylic acid on bead may first be converted to an acid chloride using oxalyl chloride, then conjugated with acceptor 73, followed removal of acetate group with NaOMe in MeOH to yield 76.

#### Scheme 13 Synthesis of Acceptors 74, 75, and 76 on Bead

## Synthesis of Test Ligands

Known sialyltransferase inhibitors, WWWWWG-NH $_2$  65 and WWWWNG-NH $_2$ , (see Lee, K. Y. et al. *J. Biol. Chem.* 2002, 277, 49341-49351) are illustrated as model inhibitors

to illustrate one-bead-one-acceptor-one-inhibitor libraries of this aspect of the present invention. As mentioned, the solid support used is not critical. Examples of suitable supports include, but are not limited to, PL-PEGA resin. As shown in Scheme 14, the resin was first coupled with Fmoc-4-NO<sub>2</sub>-

phenylaniline to reduce auto-fluorescence of the beads (see Harris, R. F. et al. *J. Am. Chem. Soc.* 2000, 122, 11270-11271). After removal of the Fmoc protecting group, the resin can be conjugated with a mixture of 4-pentynoic acid and Fmoc-trytophan (1:10). Then, the peptides WWWNG-NH $_2$  5 and WWWWG-NH $_2$  can be constructed on the solid support using standard peptide synthesis. The terminal alkyne on bead can be involved in a 1,3-dipolar cycloaddition with acceptor 68 in the presence of CuSO $_4$  and sodium ascorbate to give 77 and 78, individually (see Rostovsev, V. V. et al. *Angew.* 10 *Chem. Int. Ed.* 2002, 41, 2596-2599).

to about 2 mM and more preferably about 0.2 mM. Suitable concentrations of donor range from about 0.001 mM to about 10 mM preferably about 0.01 mM to about 1 mM and more preferably about 0.1 mM. Suitable concentrations of enzyme range from about 0.002 mU to about 20 mU preferably about 0.02 mU to about 2 mU and more preferably about 0.2 mU. Suitable concentrations of inhibitor range from about 20 µM to about 300 µM. In a typical assay, the inhibitor is first incubated with enzyme in buffer for about 30 min. Then, the donor and acceptor are added and the reaction mixture is incubated for 1 h. As a blank control, enzyme can be incu-

To test the activity and quenching efficiency of acceptors and donors, the acceptors and donors can be incubated in the presence of the enzyme, e.g. sialyltransferase. As a control, the a donor or acceptor can be incubated with the enzyme in the absence of the other.

Enzymatic assays may be carried out in a suitable buffer as is known in the art. For example, for sialyltransferases a buffer containing 62.5 mM sodium cacodylate (pH=6.0), 1 mg/mL bovine serum albumin (BSA), and 0.5% Triton X 100 may be used (see Gross, H. J. et al. *Anal. Biochem.* 1990, 186, 65 127-134) Suitable concentrations of acceptor range from about 0.002 mM to about 20 mM preferably about 0.02 mM

bated with the donor and acceptor without inhibitor. After rinsing any solid support with water, the change in the detectable label, e.g. fluorescent intensity, can be measured using a suitable device. For fluorescent labels, a fluorescence plate reader may be used. Using this competitive assay, the inhibitory activity of enzyme inhibitors can be screened.

The invention will now be described in greater detail by way of specific examples. The following examples are put forth so as to provide those of ordinary skill in the art with a complete disclosure and description of how make a detectable substrate of the present invention and to use such a substrate

to identify enzyme inhibitors, and are not intended to limit the scope of what the inventors regard as their invention.

#### **EXAMPLES**

Reagents and General Procedures. All chemicals were used as supplied without further purification. Solvents (MeOH 99.8%, CH<sub>2</sub>Cl<sub>2</sub> 99.8%, CH<sub>3</sub>CN 99.8%, DMF 99.8%) were purchased in anhydrous SURE/SEALTM bottles from Aldrich, used without further purification, and stored under 10 argon. Bacterial neuraminidases (Clostridium perfringens, Salmonella typhimurium, Vibrio cholerae) were purchased from Sigma. Sialyltransferases were purchased from Calbiochem. BSA, (Fraction V), Triton X 100, and CMP-Neu5Ac were purchase from Sigma. PL-PEGA resin (0.4 mmol/g, 15 150-300 μm) was purchased from Polymer Laboratories. Dowex 50WX8 (200 mesh) acidic resin was purchased from Aldrich, washed copiously with methanol, and used without further purification. NaOMe/MeOH (0.5 M) was purchased from Aldrich. Glass-backed EM Science thin layer chroma- 20 tography (TLC) plates (silica gel 60 with a 254 nm fluorescent indicator) were purchased from VWR International, cut into 2 cm×5 cm portions, used without further manipulation, and stored over dessicant. Developed TLC plates were visualized under a short wave UV lamp, stained with a cerium-molyb- 25 date solution and charred. Column chromatography was conducted using flash silica gel (32-63 µm) available from Scientific Adsorbents and solvents purchased from EM Science. NMR experiments (1-D and 2-D) were conducted on Bruker DRX500 MHz spectrometers at 298 K. Reverse phase-high 30 performance liquid chromatography (RP-HPLC) preparative separations were carried out on Vydac C18 column (10×250 mm). Solvents: (A) water (H<sub>2</sub>O) and (B) acetonitrile (CH<sub>3</sub>CN) with UV detection at 220 and 254 nm. Fluorescence intensity was measured using PerkinElmer fluores- 35 cence plate reader.

#### Example 1

Synthesis of Detectable Label-Substrate Conjugate 8-Allyl-4-methylumbelliferyl-4,7,8,9-tetra-O-acetyl-α-D-N-acetyl-neuraminic amide-bead conjugate

7-Allyloxy-4-methylcoumarin (2)

3-Bromo-1-propene (2.95 mL, 34.06 mmol) was added 55 dropwise under argon to a stirred mixture of 7-hydroxy-4-methylcoumarin 1 (2.0 g, 11.35 mmol) and anhydrous potassium carbonate ( $K_2CO_3$ ) (2.35 g, 17.03 mmol) in acetone (50 mL). The resulting mixture was then refluxed for 4 h, after which it was allowed to cool, and the  $K_2CO_3$  filtered off and 60 washed with fresh acetone. The solvent was removed in vacuo and the residue was crystallized from methanol to afford 2 as white-cream crystals (2.36 g, 96%). TLC (hexane/ethyl acetate, 1:3):  $R_7$ =0.63. mp 110° C.  $^1$ H NMR (CDCl $_3$ , 500 MHz)  $\delta$  2.39 (s, 3H, CH $_3$ ), 4.60 (d, J=5.0 Hz, 2H, CH $_2$ ), 5.33 65 (d, J=10.5 Hz, 1H), 5.44 (d, J=17.5 Hz, 1H), 6.05 (m, 1H), 6.12 (s, 1H), 6.80 (s, 1H), 6.87 (dd, J=1.5, 9.0 Hz, 1H), 7.49

44

(d, J=9.0 Hz, 1H).  $^{13}$ C NMR (CDCl<sub>3</sub>, 125 MHz)  $\delta$  18.70, 69.29, 101.84, 112.05, 112.79, 113.75, 118.51, 125.64, 132.31, 152.61, 152.26, 161.26, 161.65. ESIMS calcd for  $C_{13}H_{13}O_3$  [M+H] $^+$  217.1, found 217.1.

8-Allyl-7-hydroxyl-4-methyl-2H-chromen-2-one (3)

7-Allyloxy-4-methylcoumarin 2 (1.0 g, 4.62 mmol) was dissolved in N,N-diethylaniline (8 mL) in a sealed vial and heated to 250° C. for 25 min under microwave. The reaction mixture was then cooled, during which some of the product precipitated. Hexane (10 mL) was added in order to precipitate out the remaining product. The precipitate was filtered, washed with hexane, and dried under vacuum to yield 3 as a pale-yellow solid (0.88 g, 88%). TLC (hexane/ethyl acetate, 1:3): R<sub>=</sub>=0.5. mp 204° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 500 MHz) δ 2.43 (s, 3H, CH<sub>3</sub>), 3.57 (d, J=6.0 Hz, 2H, CH<sub>2</sub>), 4.95 (d, J=10.0 Hz, 1H), 5.04 (d, J=17.0 Hz, 1H), 5.97 (m, 1H), 6.10 (s, 1H), 6.86 (d, J=9.0 Hz, 1H), 7.46 (d, J=9.0 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 125 MHz) δ 18.85, 27.66, 110.78, 113.20, 113.73, 114.93, 115.41, 124.38, 136.16, 153.89, 155.88, 160.18, 163.92. ESIMS calcd for C<sub>13</sub>H<sub>13</sub>O<sub>3</sub> [M+H]<sup>+</sup> 217.1. Found 217.2.

8-Allyl-4-methylumbelliferyl-4,7,8,9-tetra-O-acetylα-D-N-acetyl-neuraminic acid methyl ester (4)

2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-α-D-N-acetyl-45 neuraminic acid methyl ester (0.68 g, 1.32 mmol) was dissolved in a mixture of 3 (0.32 g, 1.46 mmol), Ag<sub>2</sub>CO<sub>3</sub> (0.37 g, 1.32 mmol) and activated molecular sieves (1.20 g) in anhydrous acetonitrile (20 mL). The mixture was stirred under argon at room temperature in the dark overnight, filtered, and evaporated. The residue was purified by column chromatography (ethyl acetate) to give 4 as a white solid (0.71 g, 78%). TLC (ethyl acetate):  $R_{f}=0.19$ . mp 107° C.  $[\alpha]_{D}^{26}+27.0^{\circ}$  (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 1.93, 2.05, 2.12, 2.20 (4s, 15H, 4-OAc, 1-NAc), 2.32 (m, 1H, H-3a), 2.41 (s, 3H, CH<sub>3</sub>), 2.76 (dd, J=4.5, 13.0 Hz, 1H, H-3e), 3.61 (m, 5H, CO<sub>2</sub>CH<sub>2</sub>, CH<sub>2</sub>), 4.13 (m, 2H, H-5, H-9a), 4.28 (m, 1H, H-9b), 4.57 (m, 1H, H-6), 5.00 (m, 2H, H-4, CH<sub>2</sub>=CH), 5.08 (d, J=17.0 Hz, 1H, CH<sub>2</sub>=CH), 5.29 (d, J=10.0 Hz, 1H, NH), 5.39 (m, 2H, H-7, H-8), 5.93 (m, 1H, CH<sub>2</sub>=CH), 6.19 (s, 1H, C=CH), 7.17 (d, J=9.0 Hz, 1H), 7.41 (d, J=9.0 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 19.00, 20.92, 20.99, 21.22, 23.41, 27.71, 38.79, 49.71, 53.27, 62.17, 67.28, 68.61, 68.73, 73.66, 100.23, 113.38, 114.27, 115.93, 116.22, 118.67, 123.46, 135.01, 152.47, 152.78, 154.94, 161.13, 168.23, 170.16, 170.31, 170.42, 170.71, 171.06. FABHRMS calcd for C<sub>33</sub>H<sub>39</sub>NO<sub>15</sub>Na [M+Na]<sup>+</sup> 712.2212. Found 712.2222.

## Example 2

Synthesis of Detectable Label-Substrate Conjugate 4-Methyl acetate-umbelliferyl-4, 7, 8, 9-tetra-O-acetyl-α-D-N-acetyl-neuraminic amide-bead Conjugate

4-Bromomethyl-7-hydroxycoumarin (7)

NaOMe (5.38 mL, 0.5 M, 2.69 mmol) was added slowly under argon to a stirred solution of 4-bromomethyl-7-acetoxycoumarin 6 (0.80 g, 2.69 mmol) in dry methanol (15 mL). The resulting mixture was stirred for 0.5 h. The methanolic solution was acidified with Dowex 50WX8 (H+) resin, filtered, washed with methanol, and evaporated to afford 7 as a white solid (0.68 g, 100%). TLC (hexane/ethyl acetate, 1:3): R<sub>y</sub>=0.58. mp 207° C.  $^{1}$ H NMR (CD<sub>3</sub>OD, 500 MHz)  $\delta$  4.66 (s, 2H, CH<sub>2</sub>), 6.36 (s, 1H, C=CH), 6.73 (d, J=1.5 Hz, 1H), 6.84 (dd, J=1.5, 8.5 Hz, 1H), 7.71 (d, J=8.5 Hz, 1H).  $^{13}$ C NMR (CD<sub>3</sub>OD, 125 MHz)  $\delta$  27.58, 103.80, 111.17, 112.32, 114.37, 127.54, 153.61, 157.23, 163.24, 163.28. ESIMS calcd for C<sub>10</sub>H<sub>8</sub>BrO<sub>3</sub> [M+H]<sup>+</sup> 255.0. Found 255.1.

4-Bromomethylumbelliferyl-4,7,8,9-tetra-O-acetylα-D-N-acetyl-neuraminic acid methyl ester (8)

2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-α-D-N-acetylneuraminic acid methyl ester (0.55 g, 1.06 mmol) was dissolved in a mixture of  $7 (0.30 \,\mathrm{g}, 1.18 \,\mathrm{mmol})$ ,  $\mathrm{Ag_2CO_3} (0.29 \,\mathrm{g},$ 1.06 mmol) and activated molecular sieves (1.0 g) in anhydrous acetonitrile (20 mL). The mixture was stirred under argon at room temperature in the dark overnight, filtered, and 45 evaporated. The residue was purified by column chromatography (ethyl acetate) to give 8 as a white solid (0.63 g, 82%). TLC (ethyl acetate):  $R_f = 0.23$ . mp 117° C.  $[\alpha]_D^{26} + 29.7$ ° (c 0.6, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 1.93, 2.05, 2.15, 2.16 (4s, 15H, 4-OAc, 1-NAc), 2.27 (m, 1H, H-3a), 2.74 (dd, 50 J=4.5, 13.0 Hz, 1H, H-3e), 3.72 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.12 (m, 2H, H-5, H-9a), 4.30 (m, 1H, H-9b), 4.49 (s, 2H, CH<sub>2</sub>Br), 4.54 (m, 1H, H-6), 5.00 (m, 1H, H-4), 5.36 (m, 2H, H-7, H-8), 5.64 (d, J=10.0 Hz, 1H, NH), 6.45 (s, 1H, C=CH), 7.07 (d,  $J{=}1.5\,Hz,1H),7.10\,(dd,J{=}1.5,8.5\,Hz,1H),7.65\,(d,J{=}8.5\,Hz,\phantom{0}^{55}$ 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 14.30, 20.85, 20.93, 21.08, 21.13, 23.27, 26.82, 38.24, 49.44, 53.40, 60.51, 62.24, 67.52, 68.55, 69.26, 73.92, 99.97, 108.12, 113.44, 114.64, 116.12, 125.75, 149.95, 155.09, 157.10, 160.46, 168.09, 170.17, 170.22, 170.43, 170.73, 170.92. FABHRMS calcd for C<sub>30</sub>H<sub>34</sub>NO<sub>15</sub>BrNa [M+Na]<sup>+</sup> 750.1004. Found 750.1007.

## Conjugating of 8 on Resin (9)

PL-PEGA resin (20 mg, 0.4 mmol/g, 150-300 µm) was swelled in DMF overnight. Compound 8 (18 mg, 24 µmol)

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and 20  $\mu$ L DIEA were added to the resin. The resin was shaken for 8 h, and washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL).

Deprotection of 9 on Beads (10)

The resin 9 (20 mg, 0.4 mmol/g, 150-300  $\mu m$ ) was suspended in 1 mL anhydrous MeOH. NaOMe (100  $\mu L$ , 0.5M) was added and the resin was shaken for 30 min and washed with MeOH. Aqueous NaOH (500  $\mu L$ , 0.1 M) and MeOH (500  $\mu L$ ) were then added, and the resin was shaken for another 30 min, then washed with water (5×3 mL). A portion of beads was incubated with neuraminidase to confirm the conjugation.

#### Example 3

Synthesis of Detectable Label-Substrate Conjugate 4-Sodium acetate-umbelliferyl-α-D-N-acetyl-neuraminic amide-bead Conjugate

4-Methyl acetate-umbelliferyl-4,7,8,9-tetra-O-acetylα-D-N-acetyl-neuraminic acid methyl ester (13)

2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl-α-D-N-acetylneuraminic acid methyl ester (1.10 g, 2.12 mmol) was dissolved in a mixture of 12 (see Zhu, Q. et al. Organic Lett. 2003, 5, 1257-1260). (0.54 g, 2.33 mmol), Ag<sub>2</sub>CO<sub>3</sub> (0.58 g, 2.12 mmol) and activated molecular sieves (1.8 g) in anhydrous acetonitrile (30 mL). The mixture was stirred under argon at room temperature in the dark for 24 h, filtered, and evaporated. The residue was purified by column chromatog-35 raphy (ethyl acetate) to give 13 as pale yellow solid (0.66 g, 44%). TLC (ethyl acetate):  $R_f = 0.18$ . mp 105° C.  $[\alpha]_D^{26} +$ 25.6° (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 1.92, 2.04, 2.14 (3s, 15H, 4-OAc, 1-NAc), 2.25 (m, 1H, H-3a), 2.72 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.10 (m, 4H, H-5, H-9a, CH<sub>2</sub>), 4.29 (m, 1H, H-9b), 4.51 (m, 1H, H-6), 5.00 (m, 1H, H-4), 5.37 (m, 3H, H-7, H-8, NH), 6.29 (s, 1H, C=CH), 7.04 (m, 2H), 7.49 (d, J=9.0 Hz, 1H).  $^{13}{\rm C}$  NMR (CDCl $_{\!3},$  125 MHz)  $\delta$  20.86, 20.95, 21.11, 23.34, 37.92, 38.24, 49.60, 52.84, 53.45, 62.18, 67.46, 68.51, 69.19, 73.84, 99.96, 108.02, 115.15, 115.51, 116.13, 125.78, 147.83, 154.79, 156.92, 160.54, 168.06, 169.20, 170.22, 170.53, 170.79, 170.99. ESIMS calcd for C<sub>32</sub>H<sub>37</sub>O<sub>17</sub>NNa [M+Na]<sup>+</sup> 730.2. Found 730.0.

# 4-Acetic acid umbelliferyl-α-D-N-acetyl-neuraminic acid disodium salt (14)

NaOMe (0.1 mL, 0.5 M, 0.05 mmol) was added slowly under argon to a stirred solution of 13 (200 mg, 0.28 mmol) in dry methanol (10 mL). The resulting mixture was stirred for 0.5 h. The methanolic solution was acidified with Dowex 50WX8 (H<sup>+</sup>) resin, filtered, washed with methanol, and evaporated to dryness. The residue was dissolved in water (5 mL) and NaOH (5.9 mL, 0.1 M, 0.59 mmol) was added. The mixture was stirred for 1 h at room temperature, and freezedried to give a yellow solid, which was purified by RP-HPLC (H<sub>2</sub>O/CH<sub>3</sub>CN=90/10~70/30) to afford 14 as a white solid (100 mg, 64%). Decompose at 170° C.  $[\alpha]_D^{24}$ +45.3° (c 0.7, H<sub>2</sub>O). <sup>1</sup>H NMR (D<sub>2</sub>O, 500 MHz)  $\delta$  2.08 (m, 4H, NAc, H-3a),

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2.89 (dd, J=4.0, 12.5 Hz, 1H, H-3e), 3.66 (m, 2H), 3.88 (m, 6H), 4.13 (m, 1H), 6.37 (s, 1H, C=CH), 7.23 (m, 2H), 7.71 (d, J=8.5 Hz, 1H).  $^{13}$ C NMR (D<sub>2</sub>O/CD<sub>3</sub>OD, 125 MHz) 8 22.86, 42.25, 53.39, 64.0, 68.91, 69.72, 72.85, 74.91, 103.83, 109.33, 113.82, 116.55, 119.0, 127.05, 154.89, 155.05, 159.05, 164.76, 173.09, 175.87. ESIMS calcd for  $C_{22}H_{24}O_{13}NNa_2$  [M+H]+ 556.1. Found 556.1.

#### Conjugating of 14 on Beads (15)

PL-PEGA resin (40 mg, 0.4 mmol/g, 150-300  $\mu$ m) was swelled in DMF overnight. Compound 14 (27 mg, 48  $\mu$ mol), HOBt (48  $\mu$ L, 1.0 M) and DIC (8  $\mu$ L) were added to the resin. 15 The resin was shaken for 3 h (monitored by Kaiser test), and washed with DMF (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), MeOH (5×3 mL), H<sub>2</sub>O (5×3 mL).

## Example 4

Synthesis of Detectable Label-Substrate Conjugate 3-Allyloxy-6-[methyl(5-acetamido-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-α-D-glycero-D-galacto-2-nonulopyranosyl)onate]xanthene-9-spiro-1'-isobenzofuran-3'-one

2-[3-Oxo-6-(2-propenyloxy)-3H-xanthen-9-yl]-benzoic acid 2-propenyl ester (17)

3-Bromo-1-propene (0.78 mL, 9.0 mmol) was added dropwise under argon to a stirred mixture of fluorescein 16 (1.0 g, 3.0 mmol) and K<sub>2</sub>CO<sub>3</sub> (1.24 g, 9.0 mmol) in dry DMF (20 mL). The resulting mixture was stirred for 6 h at 65° C. The 50 mixture was concentrated, diluted with ethyl acetate, washed with 10% NaHCO<sub>3</sub>, brine, dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated. The residue was purified by column chromatography (ethyl acetate/hexane=1:1) to give 17 as a brown solid (0.94 g, 76%). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 4.47 (m, 2H, CH<sub>2</sub>), 4.65 (m, 2H, CH<sub>2</sub>), 5.09 (s, 1H, CH), 5.11 (d, J=5.0 Hz, 1H, CH), 5.36 (d, J=11.0 Hz, 1H, CH), 5.46 (d, J=17.0 Hz, 1H, CH), 5.59 (m, 1H, CH), 6.06 (m, 1H, CH), 6.45 (s, 1H), 6.54 (d, J=9.5 Hz, 1H), 6.76 (d, J=8.5 Hz, 1H), 6.88 (m, 2H),  $_{60}$ 6.95 (s, 1H), 7.32 (d, J=7.5 Hz, 1H), 7.68 (t, J=7.5 Hz, 1H), 7.74 (t, J=7.5 Hz, 1H), 8.26 (d, J=7.5 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz)  $\delta$  66.23, 69.65, 101.37, 105.99, 113.97, 115.19, 117.98, 118.93, 119.38, 129.09, 129.86, 130.18, 130.41, 130.70, 130.70, 130.80, 131.15, 132.09, 132.85, 65 134.60, 150.12, 154.40, 159.12, 163.17, 165.21, 185.90. ESIMS calcd for  $C_{26}H_{20}O_5Na [M+Na]^+ 435.1$ . Found 435.2.

3'-Hydroxy-6'-(2-propenyloxy)-spiro[isobenzofuran-1(3H), 9'-[9H]xanthen]-3-one (18)

NaOH (20 mL, 1.0 M, 20.0 mmol) was added to a solution of 17 (0.40 g, 0.97 mmol) in methanol (60 mL). The resulting mixture was stirred for 2 h at room temperature. The mixture was concentrated, diluted with ethyl acetate, washed with 10% NaHCO<sub>3</sub>, brine, dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, and evaporated. The residue was purified by column chromatography (ethyl acetate/hexane=1:1) to give 18 as a yellow solid (0.23 g, 65%). TLC (ethyl acetate/hexane, 3:1): R=0.6. mp 204° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 500 MHz) δ 4.58 (d, J=5.0 Hz, 2H, CH<sub>2</sub>), 5.32 (d, J=10.5 Hz, 1H, CH), 5.43 (d, J=17.5 Hz, 1H, CH), 6.04 (m, 1H, CH), 6.62 (m, 4H), 6.74 (s, 1H), 6.80 (s, 1H), 7.20 (d, J=7.5 Hz, 1H), 7.65 (t, J=7.5 Hz, 1H), 30 7.71 (t, J=7.5 Hz, 1H), 8.01 (d, J=7.5 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 125 MHz) δ 69.00, 101.71, 102.73, 109.90, 111.21, 111.94, 112.42, 117.89, 124.04, 124.80, 126.69, 128.88, 128.95, 129.72, 132.51, 135.16, 152.56, 160.28, 170.26. ESIMS calcd for  $C_{23}H_{17}O_5$  [M+H]<sup>+</sup> 373.1. Found <sup>35</sup> 373.2.

3-Allyloxy-6-[methyl(5-acetamido-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-α-D-glycero-D-galacto-2-nonulopyranosyl)onate]xanthene-9-spiro-1'-isobenzofuran-3'-one (19)

2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl- $\alpha$ -D-N-acetylneuraminic acid methyl ester (140 mg, 0.27 mmol) was dis-45 solved in a mixture of 18 (112 mg, 0.30 mmol), Ag<sub>2</sub>CO<sub>3</sub> (76 mg, 0.27 mmol) and activated molecular sieves (400 mg) in anhydrous acetonitrile (10 mL). The mixture was stirred under argon at room temperature in the dark for overnight, filtered, and evaporated. The residue was purified by column chromatography (ethyl acetate) to give 19 as a pale-yellow solid (96 mg, 42%). TLC (ethyl acetate): R<sub>f</sub>=0.26. mp 115° C.  $[\alpha]_D^{26}$ +12.9° (c 0.8, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$ 1.91, 2.03, 2.03, 2.05, 2.14 (5s, 15H, 4-OAc, 1-NAc), 2.22 (m, 1H, H-3a), 2.67 (dd, J=4.5, 13.0 Hz, 1H, H-3e), 3.70 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.06 (m, 1H, H-5), 4.16 (m, 1H, H-9a), 4.30  $(m, 1H, H-9b), 4.56 (m, 3H, H-6, CH_2—CH=CH_2), 4.99 (m, H-6, CH=CH_2), 4.90 (m, H-6, CH=CH_2), 4.90$ 1H, H-4), 5.33 (m, 3H, H-7, H-8, CH<sub>2</sub>—CH—CH<sub>2</sub>), 5.42 (d, J=17.0 Hz, 1H, CH<sub>2</sub>—CH=CH<sub>2</sub>), 5.50 (broad s, 1H, NH), 6.00 (m, 1H, CH<sub>2</sub>—CH—CH<sub>2</sub>), 6.70 (m, 5H), 7.03 (d, J=4.0 Hz, 1H), 7.19 (dd, J=7.5, 11.5 Hz, 1H), 7.62 (t, J=7.5 Hz, 1H), 7.66 (t, J=7.5 Hz, 1H), 8.02 (d, J=7.5 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 20.89, 21.00, 23.27, 37.69, 46.69, 49.45, 53.44, 62.19, 67.24, 67.79, 68.77, 69.26, 73.88, 99.98, 100.20, 102.02, 107.78, 108.13, 108.45, 111.38, 112.45, 114.77, 114.99, 118.32, 124.18, 125.23, 126.72, 128.81, 129.95, 132.73, 135.27, 152.09, 152.42, 153.26, 155.28,

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160.53, 168.01, 168.24, 169.65, 170.22, 170.36, 170.47, 171.04. FABHRMS calcd for  $C_{43}H_{43}NO_{17}Na$  [M+Na]<sup>+</sup> 868.2423. Found 868.2406.

#### Example 5

Synthesis of Detectable Label-Substrate Conjugate 3-Sodium acetate-6-(5-acetamido-3,5-dideoxy-α-D-glycero-D-galacto-2-nonulopyranosylonic acid sodium salt) xanthene-9-spiro-1'-isobenzofuran-3'-one

2-(6-Hydroxy-3-oxo-3H-xanthen-9-yl)-benzoic acid 2-propenyl ester (21)

Bromo-1-propene (0.78 mL, 9.0 mmol) was added drop- 30 wise under argon to a stirred mixture of fluorescein 16 (3.0 g, 9.0 mmol) and K<sub>2</sub>CO<sub>3</sub> (1.24 g, 9.0 mmol) in dry DMF (40 mL). The resulting mixture was stirred for 4 h at 65° C. The mixture was concentrated, diluted with ethyl acetate, washed with 10% NaHCO<sub>3</sub>, brine, dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, and 35 evaporated. The residue was purified by column chromatography (ethyl acetate/hexane=2:1) to give 21 as a brown solid (0.70 g, 21%). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$  4.42 (d, J=6.0 Hz, 2H, CH<sub>2</sub>), 5.06 (m, 2H, 2CH), 5.53 (m, 1H, CH), 6.81 (m, 2H), 6.88 (m, 2H), 6.98 (m, 2H), 7.32 (d, J=7.5 Hz, 1H), 7.68 (t, J=7.5 Hz, 1H), 7.74 (t, J=7.5 Hz, 1H), 8.26 (d, J=7.5 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 66.26, 103.90, 115.01, 119.49, 122.30, 130.02, 130.51, 130.65, 131.00, 131.38 132.77, 134.48, 155.89, 158.06, 165.08, 175.84. ESIMS calcd for C<sub>23</sub>H<sub>17</sub>O<sub>5</sub> [M+H]<sup>+</sup> 373.1. Found 373.2.

(6'-hydroxy-3-oxospiro[isobenzofuran-1(3H), 9'-[9H]xanthen]-3'-yl)oxy-acetic acid methyl ester (22)

Methyl bromoacetate (0.25 mL, 2.65 mmol) was added dropwise to a stirred mixture of 21 (0.33 g, 0.89 mmol) and

 $K_2CO_3$  (0.36 g, 2.65 mmol) in dry DMF (20 mL) under argon. The resulting mixture was stirred for 4 h at 90° C. The mixture was concentrated, diluted with ethyl acetate, washed with 10% NaHCO<sub>3</sub>, brine, dried with anhydrous Na<sub>2</sub>SO<sub>4</sub>, and 5 evaporated. The residue was dissolved in dry DCM (25 mL). Tetrakis(triphenylphosphine) palladium(0) (58 mg, 0.05 mmol) and phenylsilane (1.1 mL, 8.90 mmol) were added. The mixture was stirred for 2 h at room temperature and concentrated. The residue was purified by column chroma-10 tography (ethyl acetate/hexane=1:1) to give 22 as a yellow solid (0.22 g, 62%). TLC (ethyl acetate/hexane, 3:1): R=0.5. mp 236° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 500 MHz) δ 3.76 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.69 (s, 2H, CH<sub>2</sub>), 6.68 (m, 6H), 7.19 (d, J=7.5 Hz, 1H), 7.66 (t, J=7.5 Hz, 1H), 7.71 (t, J=7.5 Hz, 1H), 8.01 15 (d, J=7.5 Hz, 1H).  $^{13}$ C NMR (CDCl<sub>3</sub>/CD<sub>3</sub>OD, 125 MHz)  $\delta$ 52.33, 65.18, 101.86, 102.77, 109.84, 111.55, 112.37, 112.52, 124.05, 124.87, 126.67, 129.00, 129.24, 129.80, 135.21, 152.47, 152.56, 152.94, 159.34, 169.21, 170.15, 171.36, 172.03. ESIMS calcd for  $C_{23}H_{17}O_7$  [M+H]<sup>+</sup> 405.1. 20 Found 405.2.

3-Methyl acetate-6-[methyl(5-acetamido-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-α-D-glycero-D-galacto-2-nonulopyranosyl)onate]xanthene-9-spiro-1'-isoben-zofuran-3'-one (23)

2-Deoxy-2-chloro-4,7,8,9-tetra-O-acetyl- $\alpha$ -D-N-acetylneuraminic acid methyl ester (170 mg, 0.33 mmol) was dissolved in a mixture of 22 (148 mg, 0.37 mmol), Ag<sub>2</sub>CO<sub>3</sub> (92 mg, 0.33 mmol) and activated molecular sieves (400 mg) in anhydrous acetonitrile (10 mL). The mixture was stirred under argon at room temperature in the dark overnight, filtered, and evaporated. The residue was purified by column chromatography (ethyl acetate) to give 23 as a pale-yellow solid (115 mg, 40%).  $[\alpha]_D^{26}$ +12.7° (c 1.1, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 1.91, 1.96, 1.99, 2.03, 2.10 (5s, 15H, 4-OAc, 1-NAc), 2.21 (m, 1H, H-3a), 2.67 (dd, J=4.5, 13.0 Hz, 1H, H-3e), 3.73 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 3.82 (s, 3H, CO<sub>2</sub>CH<sub>3</sub>), 4.11 (m, 2H, H-5, H-9a), 4.32 (m, 1H, H-9b), 4.46 (m, 1H, H-6), 4.66 (s, 2H, CH<sub>2</sub>), 4.99 (m, 1H, H-4), 5.27 (d, J=9.5 Hz, 1H, NH), 5.34 (m, 2H, H-7, H-8), 6.70 (m, 5H), 7.03 (m, 1H), 7.19 (dd, J=7.5, 11.5 Hz, 1H), 7.62 (t, J=7.5 Hz, 1H), 7.68 (t, J=7.5 Hz, 1H), 8.02 (d, J=7.5 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 20.92, 21.01, 21.16, 23.41, 37.88, 49.68, 52.59, 53.47, 62.17, 65.51, 67.71, 68.75, 69.71, 73.87, 99.97, 100.20, 102.20, 107.86, 108.38, 112.03, 112.57, 114.68, 116.26, 124.23, 125.28, 126.74, 129.22, 129.49, 130.02, 50 135.02, 152.01, 152.47, 153.17, 155.39, 159.61, 168.22, 168.95, 169.47, 170.10, 170.17, 170.42, 170.74, 171.04. ESIMS calcd for C<sub>43</sub>H<sub>43</sub>NO<sub>19</sub>Na [M+Na]<sup>+</sup> 900.2. Found 900.1.

3-Sodium acetate-6-(5-acetamido-3,5-dideoxy-α-D-glycero-D-galacto-2-nonulopyranosylonic acid sodium salt) xanthene-9-spiro-1'-isobenzofuran-3'-one (24)

NaOMe (0.1 mL, 0.5 M, 0.05 mmol) was added slowly under argon to a stirred solution of 23 (52 mg, 0.059 mmol) in dry methanol (5 mL). The resulting mixture was stirred for 0.5 h. The methanolic solution was acidified with Dowex 50 (H<sup>+</sup>) resin, filtered, washed with methanol, and evaporated to dryness. The residue was dissolved in water (5 mL), and NaOH (1.29 mL, 0.1 M, 0.129 mmol) was added. The mixture

was stirred for 1 h at room temperature, and freeze-dried to give a yellow solid, which was purified by RP-HPLC ( $\rm H_2O/CH_3CN=90:10\sim70:30$ ) to afford 24 as pale-yellow solid (28 mg, 67%). Decompose at 150° C. [ $\rm Cl_2D^{24}+17.5^{\circ}$  (c 0.3,  $\rm H_2O$ ).  $^{1}$ H NMR (CD<sub>3</sub>OD, 500 MHz)  $\delta$  1.88 (m, 1H, H-3a), 2.02 (s, 3H, NAc), 2.97 (m, 1H, H-3e), 3.54-3.90 (m, 7H, H-4, 5, 6, 7, 8, 9a, 9b), 4.51 (s, 2H, CH<sub>2</sub>), 6.66 (m, 3H), 6.86 (m, 1H), 6.99 (m, 1H), 7.20 (m, 2H), 7.71 (t, J=7.5 Hz, 1H), 7.78 (t, J=7.5 Hz, 1H), 8.01 (d, J=7.5 Hz, 1H).  $^{13}$ C NMR (CD<sub>3</sub>OD, 125 MHz)  $\delta$  22.62, 42.89, 53.99, 64.48, 69.27, 70.27, 73.12, 75.27, 77.30, 85.06, 102.97, 110.01, 112.41, 113.20, 114.77, 118.45, 118.63, 125.27, 125.76, 127.78, 129.11, 129.83, 130.04, 131.18, 136.69, 153.06, 153.81, 154.48, 158.36,  $^{15}$  161.92, 171.54, 175.53. ESIMS calcd for  $\rm C_{33}H_{29}NO_{15}Na_3$  [M+Na]+ 748.1. Found 748.1.

#### Conjugating of 24 to Resin (25)

PL-PEGA resin (20 mg, 0.4 mmol/g, 150-300  $\mu$ m) was swelled in DMF overnight. Compound 24 (12 mg, 16  $\mu$ mol), HOBt (24  $\mu$ L, 1.0 M) and DIC (4  $\mu$ L) were added to the resin. The resin was shaken for 4 h (monitored by Kaiser test), and washed with DMF (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), MeOH (5×3 mL), H<sub>2</sub>O (5×3 mL).

#### Example 6

Synthesis of Test Ligand 5-Acetamido-4-(tert-butyl-dimethyl-silanyloxy)-8,9-O-(1-methylethylidene)-2, 6-anhydro-3,5-dideoxy-D-glycero-D-galacto-Non-2-enonic acid methyl ester (27)

5-Acetamido-4-(tert-butyldimethylsilanyloxy)-8,9-O-(1-methylethylidene)-2,6-anhydro-3,5-dideoxy-Dglycero-D-galacto-non-2-enonic acid methyl ester (27)

2,2-Dimethoxypropane (0.45 mL, 3.67 mmol) was added slowly to a stirred solution of 26 (see Kirchner, E. et al. J. Carbohydr. Chem. 1988, 7, 453-86), (187 mg, 0.61 mmol) in 45 dry acetone (10 mL) under argon. Toluenesulfonic acid (12 mg, 0.06 mmol) was added to the solution. The resulting mixture was stirred overnight at room temperature. The mixture was evaporated and dissolved in DMF (10 mL), TBSC1 (110 mg, 0.73 mmol) and imidazole (42 mg, 0.61 mmol) was 50 added. The solution was stirred under argon for 6 h at room temperature. The mixture was evaporated and extracted with ethyl acetate, and purified by column chromatography (ethyl acetate/hexane=2:1) to give 27 as a white solid (208 mg, 74%). TLC (ethyl acetate/hexane, 3:1): R=0.35. mp 245° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 0.17 (s, 3H, CH<sub>3</sub>), 0.19 (s, 3H, CH<sub>3</sub>), 0.91 (s, 9H, 3CH<sub>3</sub>), 1.37 (s, 3H, CH<sub>3</sub>), 1.41 (s, 3H, CH<sub>3</sub>), 2.04 (s, 3H, NAc), 3.60 (m, 1H), 3.81 (s, 3H, OCH<sub>3</sub>), 4.07 (m, 2H), 4.16 (t, J=8.5 Hz, 1H), 4.24 (d, J=7.0 Hz, 1H), 4.34 (m, 2H), 4.48 (m, 1H, OH), 5.54 (d, J=7.5 Hz, 1H), 5.89 (d, J=3.5 Hz, 1H, NH).  $^{13}$ C NMR (CDCl<sub>3</sub>, 125 MHz)  $\delta$  –4.46, -4.27, 18.10, 23.36, 25.46, 25.77, 27.20, 51.70, 52.63, 66.26, 67.51, 72.12, 74.63, 76.70, 109.42, 109.92, 144.97, 162.75, 171.46. ESIMS calcd for C<sub>21</sub>H<sub>37</sub>NO<sub>8</sub>SiNa [M+Na]<sup>+</sup> 482.2. 65 Found 482.3. Anal. Calc. for C<sub>21</sub>H<sub>37</sub>NO<sub>8</sub>Si: C, 54.88; H, 8.11; N, 3.05. Found: C, 54.82; H, 8.19; N, 3.03.

5-Acetylamino-4-(tert-butyldimethylsilanyloxy)-6-[(2,2-dimethyl-[1, 3]dioxolan-4-yl)-methoxycarbon-ylmethoxy-methyl]-5,6-dihydro-4H-pyran-2-car-boxylic acid methyl ester (28)

Sodium hydride (26 mg, 1.09 mmol) was added in portions to a stirred solution of 27 (250 mg, 0.54 mmol) and methyl <sub>20</sub> bromoacetate (0.1 mL, 1.09 mmol) in dry DMF (10 mL). The resulting mixture was stirred overnight at room temperature. Then, the mixture was evaporated and extracted with ethyl acetate, and purified by column chromatography (ethyl acetate/hexane=2:1) to give 28 as a white solid (205 mg, 71%). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$  0.03 (s, 3H, CH<sub>3</sub>), 0.05 (s, 3H, CH<sub>3</sub>), 0.85 (s, 9H, 3CH<sub>3</sub>), 1.27 (s, 3H, CH<sub>3</sub>), 1.41 (s, 3H, CH<sub>3</sub>), 1.91 (s, 3H, NAc), 3.49 (m, 1H, H-5), 3.70 (s, 3H, OCH<sub>3</sub>), 3.71 (s, 3H, OCH<sub>3</sub>), 4.07 (m, 2H), 4.17 (t, J=9.0 Hz, 1H), 4.26 (t, J=7.0 Hz, 1H), 4.36 (m, 1H), 4.57 (m, 1H), 4.66 (dd, J=2.0, 10.0 Hz, 1H, H-6), 5.23 (dd, J=2.0, 8.5 Hz, 1H, H-4), 5.78 (d, J=2.5 Hz, 1H, H-3), 7.02 (d, J=6.5 Hz, 1H, NH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ -4.75, -4.58, 18.04, 23.71, 24.63, 25.64, 25.81, 26.35, 51.99, 52.25, 54.0, 64.46, 64.62, 68.33, 76.07, 76.26, 78.81, 107.79, 113.61, 143.01, 162.60, 171.28, 172.39. ESIMS calcd for  $C_{24}H_{41}NO_{10}SiNa [M+Na]^{+}$ 35 554.3. Found 554.3. Anal. Calc. for  $C_{24}H_{41}NO_{10}Si: C, 54.22;$  H, 7.77; N, 2.63. Found: C, 54.14; H, 7.98; N, 2.61.

> 5-Acetylamino-4-(tert-butyldimethylsilanyloxy)-6-[carboxymethoxy-(2,2-dimethyl-[1,3]dioxolan-4-yl)methyl]-5,6-dihydro-4H-pyran-2-carboxylic acid (29)

Sodium hydroxide (5 mL, 0.1 M, 0.5 mmol) was added slowly to a stirred solution of 28 (100 mg, 0.19 mmol) in THF (5 mL). The resulting mixture was stirred for 2 h at room temperature. The organic solvent was evaporated and the aqueous solution was acidified with HCl to pH=2, extracted with ethyl acetate, and extract was concentrated to give 29 as a white solid (66 mg, 70%). mp 130° C. [ $\alpha$ ]<sub>D</sub><sup>26</sup>+35.1° (c 1.2, CH<sub>3</sub>OH). <sup>1</sup>H NMR (CD<sub>3</sub>OD, 500 MHz) 60.11 (s, 3H, CH<sub>3</sub>), 0.14 (s, 3H, CH<sub>3</sub>), 0.91 (s, 9H, 3CH<sub>3</sub>), 1.32 (s, 3H, CH<sub>3</sub>), 1.39 (s, 3H, CH<sub>3</sub>), 2.0 (s, 3H, NAc), 3.66 (s, 2H, OCH<sub>2</sub>), 3.98 (m, 2H), 4.14 (t, J=8.0 Hz, 1H), 4.20 (t, J=8.0 Hz, 1H), 4.27 (m, 1H, H-6), 4.33 (m, 1H), 4.63 (m, 1H, H-4), 5.84 (s, 1H, H-3). <sup>13</sup>C NMR (CD<sub>3</sub>OD, 125 MHz)  $\delta$  –4.65, –4.45, 18.81, 23.19, 25.43, 26.18, 26.61, 30.89, 51.97, 66.47, 68.13, 68.79, 70.76,

77.71, 78.38, 78.68, 109.35, 113.37, 144.93, 164.97, 173.24, 173.61. ESIMS calcd for  $\rm C_{22}H_{37}NO_{10}SiNa~[M+Na]^+$  526.3. Found 526.3. Anal. Calc. for  $\rm C_{22}H_{37}NO_{10}Si:~C,~52.47;~H,~7.41;~N,~2.78.~Found:~C,~52.38;~H,~7.48;~N,~2.64.$ 

5-Acetylamino-4-[2,3-bis(tert-butoxycarbonyl) guanidine]-6-(1,2,3-triacetoxy-propyl)-5,6-dihydro-4H-pyran-2-carboxylic acid methyl ester (31)

Lindlar catalyst (1 g) was added to a stirred solution of 30 (5.0 g, 11.0 mmol) in anhydrous ethanol (80 mL) and the flask was flashed with argon. Hydrogen gas was then bubbled through the vigorously stirred solution for 10 h. The reaction mixture was then filtered through celite, and the filtrate was 15 concentrated. The residue was dissolved in dry THF (50 mL). N,N'-bis(tert-butoxycarbonyl)-1H-pyrazole-1-carboxamidine (3.42 g, 11.0 mmol) and triethylamine (1.9 mL, 13.2 mmol) were then added. The mixture was stirred vigorously overnight at room temperature. The reaction mixture was then  $_{20}$ concentrated and purified by column chromatography (hexane/ethyl acetate=1:1) to give 31 as a colorless oil (6.1 g, 82%). TLC (hexane/ethyl acetate, 1:1): R<sub>z</sub>=0.20. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$  1.50 (s, 18H, 6CH<sub>3</sub>), 1.86 (s, 3H, NAc), 2.04, 2.06, 2.08 (3s, 12H, 3OAc), 3.79 (s, 3H, OCH<sub>3</sub>), 4.28 (m, 1H), 4.70 (dd, J=2.0 Hz, 10.5 Hz, 1H) 5.13 (m, 1H), 5.30 (m, 1H), 5.51 (d, J=4.0 Hz, 1H), 5.86 (d, J=1.5 Hz, 1H, H-3), 6.37 (s, 1H, NH), 7.15 (d, J=9.0 Hz, 1H), 7.69 (s, 1H), 8.51 (d, J=9.0 Hz, 1H, NH), 11.42 (s, 1H, NH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz)  $\delta$  20.80, 20.86, 20.94, 21.04, 23.00, 28.06, 28.31,  $_{30}$ 47.60, 49.19, 52.45, 60.40, 62.38, 67.89, 71.69, 77.96, 79.90, 83.90, 105.12, 109.79, 145.21, 152.69, 157.33, 161.75, 162.96, 170.22, 170.32, 170.58, 170.87, FABHRMS calcd for C<sub>29</sub>H<sub>44</sub>N<sub>4</sub>O<sub>14</sub>Na [M+Na]<sup>+</sup> 695.2751. Found 695.2750.

5-Acetylamino-4-[2,3-bis(tert-butoxycarbonyl) guanidine]-6-[(2,2-dimethyl-[1, 3]dioxolan-4-yl)hydroxy-methyl]-5,6-dihydro-4H-pyran-2-carboxylic acid methyl ester (32)

NaOMe (2 mL, 0.5 M, 1 mmol) was added to a stirred solution of 31 (3.0 g, 4.46 mmol) in anhydrous methanol (60 mL). The reaction mixture was then stirred for 30 min, Dowex 50WX8 acidic resin was added to neutralize the reaction mixture and filtered. The filtrate was concentrated and dis- 45 solved in dry acetone (50 mL). 2,2-Dimethoxypropane (4.5 mL, 36.7 mmol) and p-toluenesulfonic acid (80 mg, 0.4 mmol) were added to the reaction mixture. The resulting mixture was stirred overnight at room temperature. The reaction mixture was then concentrated and purified by column 50 chromatography (hexane/ethyl acetate=1:1) to give 32 as a white solid (2.3 g, 88%). TLC (hexane/ethyl acetate, 1:3):  $R_{z}=0.47$ . mp 126° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$  1.37 (s, 3H, CH<sub>3</sub>), 1.43 (s, 3H, CH<sub>3</sub>), 1.50 (s, 9H, 3CH<sub>3</sub>), 1.52 (s, 9H, 3CH<sub>3</sub>), 2.02 (s, 3H, NAc), 3.50 (m, 1H, H-7), 3.79 (s, 3H, 55 OCH<sub>3</sub>), 3.97 (m, 1H, H-5), 4.03 (m, 1H, H-6), 4.10 (m, 1H, H-9), 4.10 (m, 1H, H-9'), 4.40 (m, 1H, H-8), 5.15 (m, 1H, H-4), 5.28 (d, J=4.0 Hz, 1H, OH), 5.80 (d, J=2.0 Hz, 1H, H-3), 8.00 (d, J=5.5 Hz, 1H, NH), 8.64 (d, J=7.5 Hz, 1H, NH), 11.37 (s, 1H, NH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 22.98, 25.25, 60 27.12, 28.04, 28.24, 48.46 (C-4), 52.13 (OCH<sub>3</sub>), 52.41 (C-5), 67.50 (C-9), 69.75 (C-7), 74.01 (C-8), 78.53, 80.09 (C-6), 84.38, 106.67 (C-3), 109.19, 146.96 (C-2), 152.70, 157.66, 161.99, 162.35, 174.01. FABHRMS calcd for  $C_{26}H_{43}N_4O_{11}$ , [M+H]<sup>+</sup> 587.2928. Found 587.2951. Anal. Calc. for 65  $C_{26}H_{42}N_4O_{11}$ : C, 53.23; H, 7.22; N, 9.55. Found: C, 53.20; H, 7.26; N, 9.51.

5-Acetylamino-4-[2,3-bis(tert-butoxycarbonyl) guanidine]-6-[(2,2-dimethyl-[1,3]dioxolan-4-yl)-(4-nitro-phenoxycarbonyloxy)-methyl]-5,6-dihydro-4H-pyran-2-carboxylic acid methyl ester (33)

To a solution of 32 (286 mg, 0.49 mmol) in dry pyridine (10 mL) was added 4-dimethylaminopyridine (149 mg, 1.22 mmol) and 4-nitrophenylchloroformate (245 mg, 1.22 mmol). The reaction mixture was stirred vigorously overnight at room temperature. The solution was then concentrated and the residue was extracted with ethyl acetate and purified by column chromatography (hexane/ethyl acetate=1: 1) to give 33 as a white solid (271 mg, 74%). TLC (hexane/ ethyl acetate, 1:3): R<sub>e</sub>=0.69. mp 150° C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ 1.39 (s, 3H, CH<sub>3</sub>), 1.42 (s, 3H, CH<sub>3</sub>), 1.49 (s, 18H, 6CH<sub>3</sub>), 1.94 (s, 3H, NAc), 3.82 (s, 3H, OCH<sub>3</sub>), 4.14 (m, 1H, H-9), 4.23 (m, 2H, H-8, H-9'), 4.41 (m, 1H, H-5), 4.45 (m, 1H, H-6), 5.19 (t, J=9.5 Hz, 1H, H-4), 5.24 (d, J=5.0 Hz, 1H, H-7), 5.90 (d, J=2.5 Hz, 1H, H-3), 6.46 (broad s, 1H, NH), 7.53 (dd, J=1.5 Hz, 9.0 Hz, 2H, Ar), 8.26 (dd, J=1.5 Hz, 9.0 Hz, 2H, Ar), 8.58 (d, J=8.5 Hz, 1H, NH), 11.35 (s, 1H, NH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ 23.30, 25.64, 26.63, 28.20, 28.41, 48.74 (C-4), 48.89 (C-5), 52.76 (OCH<sub>3</sub>), 65.89 (C-9), 74.33 (C-7), 75.10 (C-6), 77.70 (C-8), 80.18, 84.37, 109.18 (C-2), 115.81, 122.47, 125.39, 126.39, 145.63 (C-2), 152.66, 152.86, 155.98, 157.50, 161.74, 162.88, 171.85. FABHRMS calcd for  $C_{33}H_{45}N_5O_{15}Na$  [M+Na]<sup>+</sup> 774.2810. Found 774.2848. Anal. Calc. for  $C_{33}H_{45}N_5O_{15}$ : C, 52.73; H, 6.03; N, 9.32. Found: C, 52.66; H, 6.06; N, 9.15.

#### Conjugation of 29 to Resin to Give 34 and 35

PL-PEGA resin (120 mg, 0.4 mmol/g, 150-300 μm) was swelled in DMF overnight, Fmoc-Gly-OH/Boc-Gly-OH (480 μL, 0.3 M, 9:1 mol/mol), HOBt (144 μL, 1.0 M) and DIC  $(22 \,\mu L)$  were then added to the resin. The resin was shaken for 2 h, and washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL). Orthogonally protected PL-PEGA resin (40 mg) was treated with 20% piperidine in DMF for 30 min, then washed with DMF (5×3 mL), MeOH  $(5\times3 \text{ mL})$ , CH<sub>2</sub>Cl<sub>2</sub>  $(5\times3 \text{ mL})$ , DMF  $(5\times3 \text{ mL})$ . Compound 29  $(25 \text{ mg}, 48 \mu\text{mol})$ , HOBt  $(48 \mu\text{L}, 1.0 \text{ M})$  and DIC  $(8 \mu\text{L})$  were added to the resin. The resin was shaken for 3 h, and washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF ( $5\times3$  mL). The resin was treated with TBAF (1.0 mL, 0.1 M) and shaken overnight. After washing the resin as described before, 1 mL 50% TFA in DCM was added and shaken for 30 min. After washing as described before, 20 mg resin was coupled with substrate 14 (2 mg) using HOBt (5 μL, 1.0 M) and DIC (1  $\mu$ L) to give 34. Another 20 mg resin was coupled with substrate 24 (2 mg) using HOBt (5 μL, 1.0 M) and DIC (1  $\mu$ L) to give 35.

## Conjugation of 33 to Resin to Give 36 and 37

Orthogonally protected PL-PEGA (40 mg) resin was treated with 20% piperidine in DMF for 30 min, then washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL). Compound 33 (36 mg, 48 µmol), DMAP (12 mg, 96 µmol) and pyridine (1 mL) were added to the resin. The resin was shaken for 40 h, and washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL). The resin was treated with aq. NaOH (1.0 mL, 0.1 M) and shaken for 2 h. After washing the resin as described before, 1 mL 50% TFA in DCM was added and shaken for 30 min. After washing as described before, 20 mg resin was coupled with substrate 14 (2 mg) using HOBt (5 µL, 1.0 M) and DIC (1 µL), and optionally Bu<sub>3</sub>N (5 µL) to give 36. Another 20 mg resin was coupled with substrate 24 (2 mg) using HOBt (5 µL, 1.0 M) and DIC (1 µL), and optionally Bu<sub>3</sub>N (5 µL) to give 37.

Orthogonally protected PL-PEGA resin (40 mg) was treated with 20% piperidine in DMF for 30 min, then washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), 5 DMF (5×3 mL). Ac<sub>2</sub>O (0.5 mL) and pyridine (0.5 mL) were added to the resin. The resin was shaken overnight, and washed with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL). Resin (20 mg) was coupled with substrate 14 (2 mg) using HOBt (5  $\mu$ L, 1.0 M) and DIC (1  $\mu$ L) to 10 give 38. Another 20 mg resin was coupled with substrate 24 (2 mg) using HOBt (5  $\mu$ L, 1.0 M) and DIC (1  $\mu$ L) to give 39.

## Example 7

Screening Test Ligands for Inhibition of Neuraminidase Using Solution Phase Fluorogenic Substrate Assay

To test the activity of fluorogenic substrates 15 and 25 in solution, four known inhibitors a-d (0.5 mM) were preincu-

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bated with three different neuraminidases (Clostridium perfringens, Salmonella typhimurium, and Vibrio cholerae, 5 mU, SIGMA) in 32.5 mM 2-(N-morpholino)-ethanesulfonic acid (MES) buffer (4 mM CaCl<sub>2</sub>, pH 6.5) at room temperature for 30 min individually. The substrates 15 and 25 (10 μL) were then added to the solution of each well. The fluorescence intensity was measured after 20 min. incubation using a Perkin-Elmer fluorescence plate reader with an excitation wavelength of 355 nm, emission wavelength of 460 nm for substrate 15 and an excitation wavelength of 485 nm, emission wavelength of 535 nm for substrate 25. As shown in FIG. 12, both substrate 15 and 25 had good activity for bacterial neuraminidases. These results show that the substrates 15 and 25 could be recognized by neuraminidases while still attached to beads. As shown in the IC<sub>50</sub> results in Table 1,2-deoxy-2,3-dehydro-N-acetylneuraminic acid, DANA, (b) was found to be a good inhibitor for the three bacterial neuraminidases, and 4-guanidino-Neu5Ac2en, ZAN-AMIVIR<sup>TM</sup> (d) is an inhibitor for Vibro cholerae neuramini-

TABLE 1

Solution phase IC <sub>50</sub> values				
Compound	Clostridium perfringens	Salmonella typhimurium	Vibrio cholerae	
HO AcHN HO Sialic Acid (a)	H >2 mM CO₂H	>2 mM	>2 mM	
HO OH O	4 uM CO <sub>2</sub> H	0.31 mM (0.35 mM) <sup>lit</sup>	20 uM (30 uM) <sup>lit</sup>	
HO OH OH AcHN H <sub>2</sub> N 4-Amino DANA (c)	>2 mM	>2 mM	>2 mM	
HO OH O	>2 nM CO <sub>2</sub> H	>2 mM	0.2 mM	

#### TABLE 1-continued

_Solution phase IC <sub>50</sub> values_					
Compound	Clostridium perfringens	Salmonella typhimurium	Vibrio cholerae		
H <sub>2</sub> N O O CO <sub>2</sub> H O CO <sub>2</sub> H	1.6 mM	>2 mM	>2 mM		
HO AcHN HN NH  H2N  7-Carbamide analog of Zanamivir	>2 mM	>2 mM	0.25 mM		

## Example 8

Screening Test Ligands for Inhibition of Neuraminidase Using Solid Phase Fluorogenic Substrate Assav.

To test the activity of fluorogenic substrates 15 and 25 as part of a solid-phase assay resin complexes 34-39 (20 μL) were directly incubated with three different bacterial neuraminidases (Clostridium perfringens, Salmonella typh- 40 imurium, and Vibrio cholerae, 5 mU, SIGMA) in 32.5 mM 2-(N-morpholino)-ethanesulfonic acid (MES) buffer (4 mM CaCl<sub>2</sub>, pH 6.5) at room temperature, individually. The fluorescence intensity was then measured using a fluorescence plate reader after 20 min incubation at an excitation wave- 45 length of 355 nm, emission wavelength of 460 nm for resin 34, 36, and 38, and an excitation wavelength of 485 nm. emission wavelength of 535 nm for resin 35, 37, and 39. As shown in FIG. 13, these on-bead assays gave consistent results with in-solution assays. If the testing compound has no 50 inhibition on neuraminidase, the neuraminidase could quickly cleave the  $\alpha$ -ketosidic linkage of the sialic acids from the substrates 15 and 25, and the beads showed strong fluorescence. Beads 38 and 39 as blank controls show no inhibition of three different bacterial neuraminidases, the substrates 55 attached to the same beads underwent facile cleavage the α-ketosidic linkage of sialic acid, and the beads show strong fluorescence. In contrast, beads 34 and 35 show good inhibition of three different bacterial neuraminidases, the beads show weak fluorescence intensity compared to beads 38 and 39. Beads 36 and 37 show no inhibition on neuraminidases of Clostridium perfringens and Salmonella typhimurium, the beads show as strong fluorescence intensity as beads 38 and 39. However, beads 36 and 37 show weak inhibition on Vibrio cholerae neuraminidases, the fluorescence intensity of beads 36 and 37 show weaker than beads 38 and 39, and stronger than beads 34 and 35. By measuring the fluorescence inten-

sity, the relative inhibitory activity can be easily compared. Results on both in-solution and on-bead assays confirm that the fluorogenic substrates is an efficient tool for identification of positive enzyme inhibitors. Resin functionalized 36 served 35 as a control showing no inhibition of the three bacterial neuraminidases; the NeuAc substrates attached to these beads underwent facile cleavage of the  $\alpha$ -ketosidic linkage of sialic acid resulting in strong fluorescence (FIG. 19c). In contrast, DANA functionalized beads (34) inhibit all three bacterial neuraminidases as evidenced by weak fluorescence intensity (FIG. 19a) compared to control (36). Beads possessing ZAN-AMIVIR<sup>TM</sup> show no inhibition of Clostridium perfringens and Samonella typhimurium neuraminidases and only weak inhibition of Vibrio cholerae neuraminidase (FIG. 19b), again this is completely consistent with the solution assays in Table 1 and the one-bead-one-substrate assay (FIG. 12).

## Example 9

Synthesis of Detectable Label-Donor Substrate Conjugate

3-(4-Hydroxy-3-nitro-phenyl)-2-[2-(7-hydroxy-2-oxo-2H-chromen-4-yl)-acetylamino]-propionamide (41)

Compound 41 was prepared using rink amide MBHA resin (160 mg, 0.6 mmol/g) as solid support according to standard peptide synthesis (HOBt 3 eq, DIC 3 eq) and isolated by precipitation with diethyl ether to give a yellow solid (28 mg, 68%). mp 203° C. <sup>1</sup>H NMR (DMSO, 500 MHz) δ 2.83 (m, 2H, CH<sub>2</sub>), 3.60 (s, 2H, CH<sub>2</sub>), 6.03 (s, 1H), 6.63 (m, 2H), 6.96 (d, J=8.5 Hz, 1H), 7.11 (s, 1H), 7.36 (d, J=8.5 Hz, 1H), 7.53 (s, 1H), 7.77 (d, J=1.5 Hz, 1H), 8.46 (d, J=9.0 Hz, 1H). <sup>13</sup>C NMR (DMSO, 125 MHz) δ 36.50, 40.43, 53.75, 102.26, 111.44, 111.73, 112.78, 118.89, 125.37, 126.63, 129.22,

135.92, 136.50, 151.05, 151.10, 154.94, 160.21, 161.11, 167.63, 172.65. FABHRMS calcd for  $\rm C_{20}H_{18}N_3O_8~[M\text{+}H]^+$ 428.1094. Found 428.1102.

N-[1-Carbamoyl-2-(4-hydroxy-3-nitro-phenyl)ethyl]-N'-[2-(2-{2-[2-(7-hydroxy-2-oxo-2Hchromen-4-yl)-acetylamino]-ethoxy}-ethoxy)-ethyl]succinamide (42)

Compound 42 was prepared using rink amide MBHA resin 10 (140 mg, 0.6 mmol/g) as solid support according to standard peptide synthesis (HOBt 3 eq, DIC 3 eq) and isolated by precipitation with diethyl ether to give a yellow solid (42 mg, 75%). mp 115° C. <sup>1</sup>H NMR (DMSO, 500 MHz) δ 2.24 (m, 4H, 2CH<sub>2</sub>), 2.85 (m, 2H, CH<sub>2</sub>), 3.16 (t, J=6.0 Hz, 2H, CH<sub>2</sub>), 15 3.22 (t, J=6.0 Hz, 2H, CH<sub>2</sub>), 3.37 (t, J=6.0 Hz, 2H, CH<sub>2</sub>), 3.41 (t, J=6.0 Hz, 2H, CH<sub>2</sub>), 3.50 (m, 4H, 2CH<sub>2</sub>), 3.65 (s, 2H,  $CH_2$ ), 4.34 (m, 1H), 5.73 (s, 1H), 6.16 (s, 1H), 6.72 (d, J=2.0 Hz, 1H), 6.79 (dd, J=2.0 Hz, 9.0 Hz, 1H), 7.03 (d, J=8.5 Hz, 1H), 7.41 (d, J=8.5 Hz, 1H), 7.60 (d, J=8.5 Hz, 1H), 7.77 (s, 20 34.02, 34.26, 36.26, 26.40, 41.00, 41.06, 49.27, 49.60, 67.33, 1H). <sup>13</sup>C NMR (DMSO, 125 MHz) δ 30.56, 30.69, 36.10, 38.55, 38.83, 38.89, 53.66, 66.48, 69.01, 69.14, 69.64, 102.35, 111.65, 111.81, 112.94, 118.85, 125.43, 126.84, 129.73, 136.55, 151.42, 155.10, 160.45, 161.09, 167.94, 171.94, 171.58, 171.78, 173.01. FABHRMS calcd for 25  $C_{30}H_{36}N_5O_{12}$  [M+H]<sup>+</sup> 658.2360. Found 658.2381.

Methyl (benzyl 5-(7-acetoxy-4-coumarinyl-acetamido)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-D-glyceroα-D-galacto-2-nonulopyranosid)onate (46)

Phosphorus pentachloride (1.08 g, 5.2 mmol) was added to a solution of 44 (see Ikeda, K. et al. Chem. Pharm. Bull. 1991, 39, 1305-1309), (1.51 g, 2.6 mmol) and N,N-dimethylaniline the mixture was stirred for 7 h. Then, MeOH (8 mL) was added to the solution and the mixture was stirred for another 2.5 h at the same temperature. Water (5 mL) was added and the mixture was warmed to room temperature and stirred overnight. The resulting solution was washed with aqueous 40 saturated NaHCO<sub>3</sub> and brine. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to dryness in vacuo. The residue was redissolved in dry THF (15 mL), 45 (see Gerber, S. A. et al. *Bioconjugate Chem.* 2001, 12, 603-615; and Nordlander, J. E. et al. J. Org. Chem. 1984, 49, 4107-45 4111), (2.6 mmol in 5 mL DCM) and DIEA (1.36 mL, 7.8 mmol) were added to the mixture and stirred overnight. The resulting mixture was washed with saturated NaHCO3 and brine. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to dryness. The residue was purified by col- 50 umn chromatography (hexane/ethyl acetate=1:2) to give 46 as a white solid (0.61 g, 30%). TLC (hexane/ethyl acetate, 1:3): R<sub>f</sub>=0.36. mp 88° C. [a]<sub>D</sub><sup>26</sup>+39.1° (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CHCl<sub>3</sub>, 500 MHz)  $\delta$  1.75-2.33 (m, 16H, 5OAc, H-3a), 2.65 (dd, J=4.5 Hz, 13.0 Hz, 1H, H<sub>3e</sub>), 3.59 (s, 3H, OCH<sub>3</sub>), 55 3.61(m, 2H, CH<sub>2</sub>), 4.16 (m, 3H), 4.32 (m, 1H), 4.41 (d, J=12.0 Hz, 1H), 4.82 (m, 2H), 5.32 (m, 1H), 5.46 (m, 1H), 6.23 (d, J=10.0 Hz, 1H, NH), 6.36 (s, 1H), 7.11 (m, 2H), 7.25 (m, 1H), 7.31 (m, 4H), 7.76 (d, J=8.5 Hz, 1H). <sup>13</sup>C NMR (CHCl<sub>3</sub>, 125 MHz)  $\delta$  20.67, 20.94, 20.99, 21.23, 21.26, 38.23, 40.67, 60 49.73, 52.75, 62.59, 66.96, 67.57, 68.73, 69.09, 72.27, 98.64, 110.70, 116.45, 116.93, 118.65, 126.56, 127.85, 127.94, 128.37, 137.22, 148.86, 153.57, 154.48, 160.61, 167.70, 168.37, 168.70, 170.26, 170.47, 170.77, 170.82. FABHRMS calcd for  $C_{38}H_{41}NO_{17}Na$  [M+Na]<sup>+</sup> 806.2267. Found 806.2271. Anal. Calc. for C<sub>38</sub>H<sub>41</sub>NO<sub>17</sub>: C, 58.24; H, 5.27; N, 1.79. Found: C, 58.12; H, 5.34; N, 1.76.

Methyl 5-(7-acetoxy-3,4-dihydro-4-coumarinyl-acetamido)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-D-glycero-β-D-galacto-2-nonulopyranosionate (47)

To a solution of 46 (200 mg, 0.25 mmol) in methanol (20 mL) was added palladium on charcoal (30 mg, 10% Pd). The mixture was stirred overnight under hydrogen at normal pressure. The catalyst was filtered off and washed with methanol. The filtrate was concentrated and the residue was purified by column chromatography (hexane/ethyl acetate=1:3) to give 47 as a white solid (138 mg, 78%). TLC (hexane/ethyl acetate, 1:3): R<sub>f</sub>=0.25. mp 97° C.  $[\alpha]_D^{26}$  –5.0° (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CHCl<sub>3</sub>, 500 MHz)  $\delta$  2.02-2.29 (m, 19H, 5OAc, H-3a, H-3e, CH<sub>2</sub>), 2.81 (m, 2H, CH<sub>2</sub>), 3.70 (s, 3H, OCH<sub>3</sub>), 4.03 (dd, J=8.0 Hz, 12.5 Hz, 1H, H-9a), 4.20 (m, 2H, H-5, H-9b), 4.53 (m, 1H, H-6), 5.15 (m, 1H), 5.22 (m, 1H), 5.34 (m, 1H), 6.15 (m, 1H), 6.83 (m, 2H), 7.29 (m, 1H). <sup>13</sup>C NMR (CHCl<sub>3</sub>, 125 MHz) δ 21.06, 21.24, 21.29, 31.08, 31.29, 68.37, 69.54, 71.12, 71.29, 71.56, 71.74, 95.10, 111.17, 118.01, 118.20, 123.20, 123.78, 128.39, 128.69, 150.72, 152.05, 167.34, 169.22, 169.42, 169.47, 170.05, 170.27, 170.54, 170.63, 171.19, 171.25, 171.29, 171.34. FABHRMS calcd for C<sub>31</sub>H<sub>37</sub>NO<sub>17</sub>Na [M+Na]<sup>+</sup> 718.1954. Found 718.1934. Anal. Calc. for  $C_{31}H_{37}NO_{17}$ : C, 53.53; H, 5.36; N, 2.01. Found: C, 53.42; H, 5.39; N, 2.04.

Methyl (benzyl 5-(9H-fluoren-9-ylmethoxycarbonylamino)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-D-glycero-β-D-galacto-2-nonulopyranosid)onate (48)

Phosphorus pentachloride (500 mg, 2.6 mmol) was added (1.33 mL, 10.4 mmol) in dry DCM (20 mL) at -35° C., and 35 to a solution of 44 (see Ikeda, K. et al. Chem. Pharm. Bull. 1991, 39, 1305-1309), (754 mg, 1.3 mmol) and N,N-dimethylaniline (660  $\mu$ L, 5.2 mmol) in dry DCM (15 mL) at -35° C., and the mixture was stirred for 7 h. Then, MeOH (4 mL) was added to the solution and the mixture was stirred for another 2.5 h at the same temperature. Water (2.5 mL) was added and the mixture was warmed to room temperature and stirred overnight. The resulting mixture was washed with aqueous saturated NaHCO<sub>3</sub> and brine. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to dryness in vacuo. The residue was redissolved in dry THF (10 mL), Fmoc-Cl (336 mg, 1.3 mmol) and DIEA (680 μL, 3.9 mmol) were added to the mixture and stirred overnight. The resulting mixture was washed with saturated NaHCO3 and brine. The organic phase was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to dryness. The residue was purified by column chromatography (hexane/ethyl acetate=1:1) to give 48 as a white solid (0.49 g, 52%). TLC (hexane/ethyl acetate, 1:3):  $R_{\rm p}$ =0.75. mp 92° C.  $[\alpha]_{\rm D}^{26}$ +3.1° (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CHCl<sub>3</sub>, 500 MHz)  $\delta$  1.94-2.17 (m, 13H, 4OAc, H-3a), 2.71 (dd, J=4.0 Hz, 12.5 Hz, 1H, H<sub>3e</sub>), 3.70 (s, 3H, OCH<sub>3</sub>), 3.79 (m, 1H, H-5), 4.18 (m, 4H), 4.33 (m, 2H), 4.44 (d, J=12.0 Hz,1H), 4.63 (d, J=10.0 Hz, 1H, NH), 4.83 (d, J=12.0 Hz, 1H), 4.93 (m, 1H, H-4), 5.48 (s, 2H, OCH<sub>2</sub>), 7.33 (m, 7H), 7.39 (t, J=7.5 Hz, 2H), 7.58 (m, 2H), 7.75 (d, J=7.5 Hz, 2H). <sup>13</sup>C NMR (CHCl<sub>3</sub>, 125 MHz)  $\delta$  21.03, 21.09, 21.42, 38.56, 47.28, 51.78, 52.91, 62.58, 67.15, 67.35, 67.63, 67.72, 68.65, 69.20, 72.81, 98.76, 120.20, 120.24, 125.25, 125.55, 127.32, 127.90, 127.97, 128.11, 128.50, 141.48, 141.59, 143.79, 156.08, 168.60, 170.32, 170.95. FABHRMS calcd for  $C_{40}H_{43}NO_{14}Na$  [M+Na]<sup>+</sup> 784.2576. Found 784.2616. Anal. Calc. for C<sub>40</sub>H<sub>43</sub>NO<sub>14</sub>: C, 63.07; H, 5.69; N, 1.84. Found: C, 63.02; H, 5.78; N, 1.82.

Methyl (4-methoxybenzyl 5-(7-acetoxy-4-coumarinylacetamido)-4,7,8,9-tetra-O-acetyl-3,5-dideoxy-Dglycero-α-D-galacto-2-nonulopyranosid)onate (50)

Phosphorus pentachloride (70 mg, 0.34 nmol) was added 5 to a solution of 49 (see Ikeda, K. et al. Bioorg. Med. Chem. Lett. 2002, 12, 2309-2311), (103 mg, 0.17 mmol) and N,Ndimethylaniline (86 µL, 0.67 mmol) in dry DCM (5 mL) at -35° C., and the mixture was stirred for 7 h. Then, MeOH (2 mL) was added to the solution and the mixture was stirred for 10 another 2.5 h at the same temperature. Water (1 mL) was added and the mixture was warmed to room temperature and stirred overnight. The resulting mixture was washed with aqueous saturated NaHCO<sub>3</sub> and brine. The organic phase was dried over anhydrate Na<sub>2</sub>SO<sub>4</sub> and evaporated to dryness in 15 vacuo. The residue was redissolved in dry THF (5 mL), 45 (see Gerber, S. A. et al. Bioconjugate Chem. 2001, 12, 603-615); and Nordlander, J. E. et al. J. Org. Chem. 1984, 49, 4107-4111), (0.17 mmol in 1 mL DCM) and DIEA (68 μL, 0.39 mmol) were added to the mixture and stirred for over- 20 night. The resulting mixture was washed with saturated NaHCO<sub>3</sub> and brine. The organic phase was dried over anhydrous Na2SO4 and evaporated to dryness. The residue was purified by column chromatography (hexane/ethyl acetate=1: 3) to give 50 as a white solid (38 mg, 28%). TLC (hexane/ 25 ethyl acetate, 1:3):  $R_{j}$ =0.29. [ $\alpha$ ]<sub>D</sub><sup>26</sup>+16.1° (c 0.4, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CHCl<sub>3</sub>, 500 MHz)  $\delta$  1.70-2.33 (m, 16H, 5OAc, H-3a), 2.61 (dd, J=4.5 Hz, 13.0 Hz, 1H, H<sub>3e</sub>), 3.53 (m, 2H, CH<sub>2</sub>) 3.68 (s, 3H, OCH<sub>3</sub>), 3.79 (s, 3H, OCH<sub>3</sub>), 4.02 (m, 1H), 4.14 (m, 1H), 4.33 (d, J=11.5 Hz, 1H), 4.73 (d, J=11.5 Hz, 1H), 4.82 30 (m, 1H), 5.29 (m, 1H), 5.46 (m, 2H), 6.34 (s, 1H), 6.85 (d, J=8.5 Hz, 2H), 7.10 (dd, J=2.5 Hz, 9.0 Hz, 1H), 7.14 (d, J=2.5 Hz, 1H), 7.24 (d, J=8.5 Hz, 2H), 7.69 (d, J=9.0 Hz, 1H). <sup>13</sup>C NMR (CHCl<sub>3</sub>, 125 MHz)  $\delta$  20.80, 21.04, 21.12, 21.35, 21.41,

## Methyl 5-(7-acetoxy-4-coumarinyl-acetamido)-4,7,8, 9-tetra-O-acetyl-3,5-dideoxy-D-glycero-α-D-galacto-2-nonulopyranosonate (51)

68.89, 7.46, 98.73, 110.89, 113.96, 116.74, 116.95, 118.70,

126.45, 129.82, 148.36, 153.55, 154.75, 160.20, 167.66,

168.53, 168.77, 170.37, 170.67, 170.89, 171.08. ESIMS calcd for C<sub>39</sub>H<sub>43</sub>NO<sub>18</sub>Na [M+Na]<sup>+</sup> 836.2, found 836.2. Anal.

57.50; H, 5.48; N, 1.74.

To a solution of 50 (20 mg, 0.025 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3 mL) and  $H_2O$  (167  $\mu$ L) was added DDQ (8.4 mg, 0.037 mmol). The mixture was stirred for 40 h at room temperature. The with anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated. The residue was purified by column chromatography (hexane/ethyl acetate=1: 3) to give 51 as a white solid (13.6 mg, 80%). TLC (hexane/ ethyl acetate, 1:3):  $R_{\epsilon}=0.20$ . mp 105° C.  $[\alpha]_{D}^{26}+7.6^{\circ}$  (c 0.5, CHCl<sub>3</sub>). <sup>1</sup>H NMR (ČHCl<sub>3</sub>, 500 MHz) δ 1.78-2.33 (m, 17H, 55 5OAc, H-3a, H-3e), 3.63 (m, 2H, CH<sub>2</sub>), 3.85 (s, 3H, OCH<sub>3</sub>), 4.02 (m, 1H), 4.20 (m, 1H), 4.47 (m, 2H), 5.19 (m, 2H), 5.30 (m, 1H), 6.09 (d, J=10.0 Hz, 1H, NH), 6.38 (s, 1H), 7.09 (dd, 1H)J=2.5 Hz, 9.0 Hz, 1H), 7.14 (d, J=2.5 Hz, 1H), 7.73 (d, J=9.0 Hz, 1H). <sup>13</sup>C NMR (CHCl<sub>3</sub>, 125 MHz) δ 20.86, 21.05, 21.23, 60 21.36, 36.32, 40.55, 50.21, 53.77, 62.87, 68.16, 69.27, 70.87, 71.39, 95.06, 110.80, 116.61, 117.11, 118.61, 126.54, 148.73, 153.63, 154.72, 160.55, 167.80, 168.80, 169.18, 170.63, 171.08, 171.14, 171.23. ESIMS calcd for C<sub>31</sub>H<sub>35</sub>NO<sub>17</sub>Na [M+Na]<sup>+</sup> 716.2. Found 716.4. Anal. Calc. 65 for C<sub>31</sub>H<sub>35</sub>NO<sub>17</sub>: C, 53.68; H, 5.09; N, 2.02. Found: C, 53.52; H, 5.16; N, 2.04.

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CMP-5-(7-hydroxy-4-coumarinyl-acetamido)-NeuAc (55)

To a solution of 7-hydroxy-4-coumarinylacetic acid N-succinimidyl ester (see Demant, E. J. F. Biochimica et Biophysica Acta 1996, 1304, 43-55), (0.5 mg, 1.6 μmol) in DMSO (0.3 mL) was added a solution of 58. Dufner, G. et al. Eur. J. Org. Chem. 2000, 1467-1482. (1 mg, 1.6 μmol) in H<sub>2</sub>O (0.2 mL). Then, a solution of saturated NaHCO<sub>3</sub> was added to control the pH 8.59.0. The mixture was stirred for 1 h at room temperature. The mixture was purified by RP-HPLC (H<sub>2</sub>O/  $CH_3OH=90:10\sim70:30$ ) to give 55 as a white powder (0.4 mg, 32%).  ${}^{1}$ H NMR (D<sub>2</sub>O, 500 MHz)  $\delta$  1.66 (m, 1H, H-3a), 2.50 (dd, J=4.5 Hz, 13.0 Hz, 1H, H-3e), 3.41 (d, J=10.0 Hz, 1H), 3.54 (dd, J=7.0 Hz, 12.0 Hz, 1H), 3.77 (s, 1H), 3.83 (dd, J=2.0 Hz, 2H), 3.83 (dd, J=2.0 Hz, 2Hz, 11.5 Hz, 1H), 3.93 (m, 5H), 4.08 (m, 1H), 4.19 (d, J=10.5 Hz, 1H), 4.25 (m, 3H), 4.31 (t, J=4.5 Hz, 1H), 4.35 (m, 1H), 5.99 (d, J=4.5 Hz, 1H), 6.10 (d, J=7.5 Hz, 1H), 6.24 (s, 1H), 6.75 (d, J=2.0 Hz, 1H), 6.84 (dd, J=2.0 Hz, 9.0 Hz, 1H), 7.57 (d, J=9.0 Hz, 1H), 7.97 (d, J=7.5 Hz, 1H).  $^{13}$ C NMR (D<sub>2</sub>O, 125 MHz) 8 38.97, 41.25, 43.07, 52.17, 63.21, 65.01, 66.81, 66.91, 69.07, 69.50, 69.89, 71.87, 74.53, 83.09, 89.34, 96.80, 100.28, 103.84, 110.71, 115.77, 126.46, 141.79, 152.13, 155.66, 158.00, 165.10, 166.39, 171.98, 172.37, 174.51. ESIMS calcd for  $C_{31}H_{38}N_5O_{20}P$  [M-H]<sup>-</sup> 830.2. Found 830.2.

## Example 10

Synthesis of Detectable Label-Donor Substrate Conjugate CMP-5-fluoresceinylaminoacetyl-Neu (59)

To a solution of fluoresceinyl isothiocyanate (12 mg, 30.8 32.19, 40.67, 50.38, 52.98, 55.54, 62.60, 66.96, 67.66, 68.77, 35 µmol) in DMF (1 mL) was added a solution of 58 (see Dufner, G. et al. Eur. J. Org. Chem. 2000, 1467-1482), (10 mg, 16 μmol) in H<sub>2</sub>O (1 mL). Then, a solution of saturated NaHCO<sub>3</sub> was added to control the pH 8.5~9.0. The mixture was stirred for 30 min at room temperature. The mixture was purified by RP-HPLC (H<sub>2</sub>O/CH<sub>3</sub>OH=90:10~60:40) to give 59 as brown Calc. for C<sub>39</sub>H<sub>43</sub>NO<sub>18</sub>: C, 57.56; H, 5.33; N, 1.72. Found: C, 40 powder (15 mg, 92%).  $[\alpha]_D^{26} + 8.1^{\circ}$  (c 1.0, H<sub>2</sub>O) <sup>1</sup>HNMR  $(D_2O, 500 \text{ MHz}) \delta 1.63 \text{ (m, 1H, H-3a)}, 2.47 \text{ (dd, J=4.5 Hz,})$ 13.0 Hz, 1H, H-3e), 2.81 (s, 1H), 2.97 (s, 1H), 3.59 (m, 1H), 3.70 (m, 1H), 3.90 (m, 2H), 4.01 (t, J=10.0 Hz, 1H), 4.13 (m, 45 1H), 4.18 (m, 7H), 5.93 (d, J=4.5 Hz, 1H), 6.05 (d, J=7.5 Hz, 1H), 6.54 (m, 3H), 7.16 (m, 3H), 7.55 (d, J=8.0 Hz, 1H), 7.72 (s, 1H), 7.89 (m, 1H).  $^{13}$ C NMR (D<sub>2</sub>O, 125 MHz)  $\delta$  37.10, 41.23, 47.81, 52.15, 63.46, 64.98, 66.92, 69.14, 69.52, 70.13, 71.88, 74.41, 83.06, 89.35, 96.78, 100.32, 103.84, 112.57, mixture was washed with saturated NaHCO<sub>3</sub>, brine, dried 50 123.06, 124.78, 126.00, 129.77, 131.44, 131.62, 141.19, 141.74, 157.94, 158.41, 158.86, 160.84, 165.12, 166.31, 172.62, 174.37, 174.58, 180.70. ESIMS calcd for  $C_{41}H_{43}N_6O_{21}PS [M-H]^- 1017.2$ . Found 1017.1.

## Example 11

Synthesis of Detectable Label-Donor Substrate Conjugate CMP-5-(7-amino-4-methylcoumarinyl-acetamido)-NeuAc (60)

To a solution of 7-amino-4-methyl-3-coumarinylacetic acid N-succinimidyl ester (see Stefanova, H. I. et al. Biochem. 1993, 32, 356-62), (0.53 mg, 1.6 μmol) in DMSO (0.3 mL) was added a solution of 58 (see Dufner, G. et al. Eur. J. Org. Chem. 2000, 1467-1482), (1 mg, 1.6 μmol) in H<sub>2</sub>O (0.2 mL). Then, a solution of saturated NaHCO3 was added to control the pH 8.5~9.0. The mixture was stirred for 1 h at room

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temperature. The mixture was purified by RP-HPLC ( $\rm H_2O/CH_3OH=90:10\sim70:30$ ) to give 60 as a white powder (0.5 mg, 36%).  $^{1}\rm H$  NMR ( $\rm D_2O$ , 500 MHz)  $\delta$  1.67 (m, 1H, H-3a), 2.43 (s, 3H, CH<sub>3</sub>), 2.51 (dd, J=4.5 Hz, 13.0 Hz, 1H, H-3e), 3.42 (d, J=10.0 Hz, 1H), 3.58-3.77 (m, 3H), 3.87 (m, 1H), 3.93-4.02 5 (m, 5H), 4.20-4.32 (m, 6H), 4.39 (t, J=5.0 Hz, 1H), 5.96 (d, J=4.5 Hz, 1H), 6.01 (d, J=7.5 Hz, 1H), 6.69 (s, 1H), 6.82 (d, J=8.5 Hz, 1H), 7.62 (m, 1H), 7.89 (d, J=7.5 Hz, 1H).  $^{13}\rm C$  NMR ( $\rm D_2O$ , 125 MHz)  $\delta$  15.00, 34.15, 43.04, 52.18, 63.27, 65.01, 66.78, 69.13, 69.64, 69.96, 71.90, 74.54, 83.12, 89.24, 10 96.66, 100.31, 112.02, 113.47, 113.67, 127.13, 141.63, 151.58, 153.98, 166.23, 172.29, 174.50. ESIMS calcd for  $\rm C_{32}H_{41}N_6O_{19}\rm P$  [M-H]  $^{-}$  843.2. Found 843.4.

#### Example 12

Synthesis of Quencher-Acceptor Substrate Conjugate 2-(3-Nitro-L-tyrosyl-amide)ethyl β-D-galactopyransyl-(1→4)-O-β-D-glucopyranoside (64)

To a solution of 63 (see Hatanaka, Y. et al. Bioorg. Med. Chem. Lett. 1995, 5, 2859-2862), (24 mg, 35 µmol) in THF (5 mL) was added Nα-Fmoc-Tyr(3-NO<sub>2</sub>)-OPfp (44 mg, 71  $\mu$ mol) and DIEA (12  $\mu$ L, 71  $\mu$ mol). The mixture was stirred for overnight at room temperature. The mixture was concen- 25 trated and the residue was purified by column chromatography (ethyl acetate). The purified compound was dissolved in 20% piperidine in DMF (2 mL), and stirred for 30 min. The mixture was concentrated and the residue was redissolved in dry methanol (2 mL). NaOMe (0.2 mL, 0.5 M) was added to 30 the solution. The mixture was stirred for 1 h, then neutralized with Dowex 50WX8 (H+) resin, filtered, washed with methanol, and evaporated. The residue was purified by RP-HPLC  $(H_2O/CH_3OH=90:10\sim80:20)$  to give 64 as a yellow solid (4.2) mg, 20%). <sup>1</sup>H NMR (D<sub>2</sub>O, 500 MHz) δ 3.17-3.28 (m, 4H), 35 3.37-3.39 (m, 2H), 3.55 (m, 2H), 3.61-3.81 (m, 8H), 3.93 (m, 2H), 4.17 (m, 1H), 4.42 (m, 2H), 7.20 (d, J=8.5 Hz, 1H), 7.55 (d, J=8.5 Hz, 1H), 8.02 (s, 1H).  $^{13}$ C NMR (D<sub>2</sub>O, 125 MHz)  $\delta$ 39.52, 42.75, 54.32, 54.56, 60.24, 61.24, 68.20, 68.73, 71.14,  $72.72,\ 74.53,\ 74.95,\ 75.58,\ 78.57,\ 102.27,\ 103.15,\ 120.57,\ \ 40$ 126.18, 126.53, 134.32, 138.64, 153.12, 168.82. ESIMS calcd for  $C_{23}H_{36}N_3O_{15}$  [M+H]<sup>+</sup> 594.2. Found 594.1.

### Example 13

Synthesis of Quencher-Acceptor Substrate Conjugate 2-(3-Nitro-L-tyrosyl-amide)ethyl β-D-galacto-pyransyl-(1→4)-O-2-acetamido-2-deoxy-β-D-glucopyranoside (68)

To a solution of 67 (see Mohan, H. et al. Synlett 2003, 9, 1255-1256), (6 mg, 8.5 μmol) in methanol (2 mL) was added palladium on charcoal (5 mg, 10% Pd). The mixture was stirred 4 h under hydrogen at normal pressure. The catalyst was filtered off and washed with methanol. The filtrate was 55 concentrated and redissolved in THF (1 mL), Nα-Fmoc-Tyr  $(3-NO_2)$ -OPfP (8 mg, 12.9 µmol) and DIEA (3 µL, 18 µmol) was added. The mixture was stirred overnight at room temperature. The mixture was concentrated and the residue was purified by column chromatography (ethyl acetate, R,=0.12). 60 The purified compound was dissolved in 20% piperidine in DMF (1 mL), and stirred for 30 min. The mixture was concentrated and the residue was redissolved in dry methanol (1 mL). NaOMe (0.1 mL, 0.5 M) was added to the solution. The mixture was stirred for 1 h, then neutralized with Dowex 65 50WX8 (H+) resin, filtered, washed with methanol, and evaporated. The residue was purified by RP-HPLC (H<sub>2</sub>O/

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CH<sub>3</sub>OH=90:10~80:20) to give 68 as a yellow solid (0.76 mg, 14%).  $^{1}$ H NMR (D<sub>2</sub>O, 500 MHz)  $\delta$  1.99 (s, 3H, NAc), 3.15 (m, 2H), 3.23 (m, 2H), 3.41 (m, 1H), 3.55 (m, 1H), 3.60 (m, 1H), 3.64-3.77 (m, 8H), 3.82 (dd, J=5.0 Hz, 12.5 Hz, 1H), 3.93 (m, 2H), 4.15 (t, J=8.0 Hz, 1H), 4.46 (d, J=7.5 Hz, 2H), 7.20 (d, J=8.5 Hz, 1H), 7.53 (dd, J=2.0 Hz, 8.5 Hz, 1H), 8.02 (d, J=2.0 Hz, 1H).  $^{13}$ C NMR (D<sub>2</sub>O, 125 MHz)  $\delta$  39.62, 44.75, 54.27, 55.13, 60.24, 61.25, 67.97, 68.75, 71.16, 72.65, 72.72, 74.93, 75.60, 78.63, 101.45, 103.11, 120.66, 126.19, 126.45, 138.65, 153.17, 168.76, 174.84. ESIMS calcd for  $C_{25}H_{39}N_4O_{15}$  [M+H] $^+$  635.23. Found 635.15.

#### Example 14

Synthesis of Quencher-Acceptor Substrate Conjugate 4-Aminophenyl 2,3,4,6-tetra-O-acetyl-β-D-galactopyransyl-(1→3)-O-2-acetamido-2-deoxy-β-D-glucopyranoside (73)

To a solution of 72 (see Matta, K. L. and Barlow, J. J. Carbohydr. Res. 1975, 43, 299-304), (46 mg, 60.5 µmol) in methanol (5 mL) was added palladium on charcoal (10 mg, 10% Pd). The mixture was stirred overnight under hydrogen at normal pressure. The catalyst was filtered off and washed with methanol. The filtrate was concentrated and the residue was purified by column chromatography (ethyl acetate/ methanol=6:1) to give 73 as a white solid (16.3 mg, 42%). TLC (methanol/ethyl acetate, 1:6): R<sub>2</sub>=0.20. mp 136° C.  $[\alpha]_D^{26}$ +12.9° (c 0.7, CH<sub>3</sub>OH). <sup>1</sup>H NMR (CD<sub>3</sub>OD, 500 MHz) δ 1.85, 1.92, 1.94, 1.99, 2.06 (5s, 1SH, 4OAc, 1NAc), 3.09 (m, 1H), 3.32 (m, 1H), 3.39 (t, J=9.5 Hz, 1H), 3.64 (m, 1H), 3.81 (m, 1H), 4.08 (m, 3H), 4.68 (d, J=7.0 Hz, 1H), 4.75 (d, J=7.0 Hz, 1H), 5.05 (m, 2H), 5.30 (m, 1H), 6.57 (d, J=8.5 Hz, 1H), 6.73 (d, J=8.5 Hz, 1H).  $^{13}$ C NMR (CD<sub>3</sub>OD, 125 MHz) δ 20.48, 20.56, 20.88, 23.26, 47.88, 56.21, 62.55, 68.75, 70.01, 70.22, 72.14, 72.40, 77.78, 84.60, 102.29, 102.41, 117.67, 119.16, 143.88, 152.23, 171.42, 171.45, 171.89, 172.09, 173.41. ESIMS calcd for  $C_{28}H_{39}N_2O_{15}$  [M+H]<sup>+</sup> 643.2. Found 643.1.

#### Example 15

Testing Acceptors 64 and 68 for Substrate Recognition by Sialyltransferases

To test the activity and quenching efficiency of acceptors with fluorescent donors, the acceptors 64 and 69 were chosen to incubate with the fluorescent donor 55 in the presence of sialyltransferase. As a control, the fluorescent donor 55 was also incubated in two blank wells plated with 100 µL sodium cacodylate buffer (pH 6.0) and sialyltransferase in the absence of acceptor. The enzymatic reaction was carried out in a buffer containing 62.5 mM sodium cacodylate (pH=6.0), 1 mg/mL BSA, and 0.5% Triton X 100 (see Gross, H. J. et al. Anal. Biochem. 1990, 186, 127-134). The plate was immediately incubated at 37° C. and 1000 rpm in a Jitterburg<sup>TM</sup> microplate incubator (Boekel). The fluorescence intensity of each well was measured using a PerkinElmer fluorescence plate reader every 20 min in the first 2 h and every 60 min in the next 2 h at an excitation wavelength of 355 nm, emission wavelength of 460 nm. As shown in FIG. 15, under normal incubation conditions (acceptor 0.2 mM, donor 0.1 mM, sialyltransferase 0.2 mU), acceptor 69 and donor 55 are good substrates for both  $\alpha$ 2,3-sialyltransferase and  $\alpha$ 2,6-sialyltransferase. Acceptor 64 showed poor substrate recognition by both enzymes. This is consistent results with literature report (see Limberg, G. et al. Liebigs Ann. 1996, 1773-1784). In the first 20% consumption of donor 55, the change of fluorescence intensity gave very good linear response with incubation time on both enzymes. These assays showed that the fluorescent donor 55 and the quenching acceptor 69 are good FRET pair for sialyltransferase assay.

#### Example 16

## Sialyltransferase Competitive Assay

To test the inhibitory activity of potential inhibitors of sialyltransferases, CMP-Neu5Ac was used as a model testing compound. Various concentrations of CMP-Neu5Ac (0 µM,  $20 \mu M$ ,  $50 \mu M$ ,  $100 \mu M$ ,  $300 \mu M$ ) were first incubated with α2,6-sialyltransferase (0.2 mU) in sodium cacodylate buffer (pH=6.0) for 30 min. Then, the fluorescent donor 55 (0.1 mM) and quenching acceptor 69 (0.2 mM) were added and the reaction mixtures were incubated at 37° C. and 1000 rpm in a  $\label{eq:linear_microplate} \mbox{ Jitterburg}^{\mbox{\scriptsize TM}} \mbox{ microplate incubator} \mbox{ (Boekel) for 1 h. As a blank}$ control, a2,6-sialyltransferase (0.2 mU) was also incubated with the fluorescent donor 55 (0.1 mM) and quenching acceptor 69 (0.2 mM) without CMP-Neu5Ac for 1 h. The fluorescence intensity of each well was measured using a PerkinElmer fluorescence plate reader at an excitation wavelength of 355 nm, emission wavelength of 460 nm. As shown in FIG. 16, fluorescence intensities showed continual increase with increase of CMP-Neu5Ac concentrations and FIG. 17 illustrates the recognition of on-bead acceptors by sialyltransferases. If one considers the blank as 100% no inhibition, the  $IC_{50}$  of CMP-Neu5Ac for  $\alpha$ 2,6-sialyltransferase is 85  $\mu$ M, which is consistent with a KM value of 45  $\mu$ M (see Gross, H. J. et al. Biochem. 1989, 28, 7386-7392). Using this competitive assay, one can simply test the inhibitory activity of sialyltransferase inhibitors.

#### Example 17

## Synthesis of On-Bead Acceptors 74, 75, and 76.

PL-PEGA resin (60 mg, 0.4 mmol/g, 150-300 μm) was swelled in DMF overnight. The resin was coupled with Fmoc-40 Phe(4-NO<sub>2</sub>)—OH (3 eq) using HOBt (3 eq) and DIC (3 eq) for 3 h. After rinsing resin with DMF (5×3 mL), MeOH (5×3 mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL), the Fmoc protecting group was cleaved by incubation with 20% piperidine in DMF for 30 min. After rinsing resin as described before, the 45 resin was distributed into two portions (20 mg and 40 mg). One portion (40 mg) of the resin was coupled with a mixture of Fmoc/Ac-Gly-OH (1:10, 3 eq) using HOBt (3 eq) and DIC (3 eq) for 3 h. After rinsing resin as described before, the Fmoc protecting group was cleaved with 20% piperidine in 50 DMF for 30 min, the resin was then reacted with succinic anhydride (10 eq) in the presence of DIEA (10 eq) in DMF for 3 h. After rinsing resin as described before, the resin was equally distributed to two portions. One portion was coupled with acceptor 64 (3 eq) using HOBt (3 eq) and DIC (3 eq) for 55 3 h, followed by removal of acetate group using NaOMe (0.1 mL, 0.5 M) in MeOH (1 mL) for 1 h to give 74. Another portion was activated to acid chloride using oxalyl chloride (0.2 mL) in DCM (1 mL) for 2 h, then reacted with acceptor 73 (3 eq) in the presence of DIEA (10 eq) in DMF for 4 h to 60 give 76. The other portion (20 mg) of resin was coupled with a mixture of 4-pentynoic acid/Ac-Gly-OH (1:10, 3 eq) using HOBt (3 eq) and DIC (3 eq) in DMF for 3 h. After rinsing resin as described before, the resin was reacted with acceptor 68 (2 eq) in the presence CuSO<sub>4</sub> (1 eq) and sodium ascorbate 65 (1 eq) in 'BuOH (0.4 mL) and H<sub>2</sub>O (0.4 mL) overnight to afford 75.

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## Example 18

Testing of On-Bead Acceptors 74, 75 and 76 for Substrate Recognition by sialyltransferases

The beads 76, 77 and 78 (20  $\mu$ L) were each plated into two wells containing 100  $\mu$ L sodium cacodylate buffer (pH 6.0). Fluorescent donor 59 (10  $\mu$ L, 1 mM) was added to each well, followed by addition of sialyltransferase (0.1 mU). The plate was immediately incubated at 37° C. and 1000 rpm in a Jitterburg<sup>TM</sup> microplate incubator (Boekel) for 6 h. The beads in each well were rinsed with water (200×5  $\mu$ L) to remove the remaining fluorescent donor 59. After washing, sodium cacodylate buffer (100  $\mu$ L) was added to each well. The fluorescence intensity of each well was measured using a PerkinElmer fluorescence plate reader at an excitation wavelength of 485 nm, emission wavelength of 535 nm.

## Example 19

## Synthesis of Test Ligands 77 and 78

PL-PEGA resin (40 mg, 0.4 mmol/g, 150-300 µm) was swelled in DMF overnight. The resin was coupled with Fmoc-Phe(4-NO<sub>2</sub>)—OH (3 eq) using HOBt (3 eq) and DIC (3 eq) for 3 h. After rinsing resin with DMF ( $5\times3$  mL), MeOH ( $5\times3$ mL), CH<sub>2</sub>Cl<sub>2</sub> (5×3 mL), DMF (5×3 mL), the Fmoc protecting group was cleaved by incubation with 20% piperidine in DMF for 30 min. After rinsing resin as described before, the resin was coupled with a mixture of 4-pentynoic acid/Fmoc-Trp-OH (1:10, 3 eq) using HOBt (3 eq) and DIC (3 eq) in DMF for 3 h. After rinsing resin as described before, the resin was equally distributed to two portions. One portion was constructed with peptide WWWWNG-NH<sub>2</sub>, another portion was constructed with peptide WWWWWG-NH<sub>2</sub> using standard peptide synthesis. Then, the resin was reacted with acceptor 68 (2 eq) in the presence CuSO<sub>4</sub> (1 eq) and sodium ascorbate (1 eq) in BuOH (0.4 mL) and H<sub>2</sub>O (0.4 mL) overnight to give 77 and 78, respectively.

## Example 20

Screening Test Ligands for Inhibition of Sialyltransferase Using Solid Phase Fluorogenic Substrate Assay.

The beads 77, 78 and 75 (20  $\mu$ L) were plated into two wells containing 100  $\mu$ L sodium cacodylate buffer (pH 6.0). Fluorescent donor 59 (10  $\mu$ L, 1 mM) and sialyltransferase (0.1 mU) were added to each well. The plate was immediately incubated at 37° C. and 1000 rpm in a Jitterburg<sup>TM</sup> microplate incubator (Boekel) for 6 h. The beads in each well were rinsed with water (200×5  $\mu$ L) to remove the remaining fluorescent donor 59. After washing, sodium cacodylate buffer (100 mL) was added to each well. The fluorescence intensity of each well was measured using a PerkinElmer fluorescence plate reader at an excitation wavelength of 485 nm, emission wavelength of 535 nm. FIG. 18 illustrates the inhibitory activity of 75, 77 and 78 against sialyltransferases.

The neuraminidase inhibitors identified using the methods of the present invention will be especially effective in treating influenza. Examples of neuraminidase inhibitors which are employed in compositions known in the art to treat influenza include, but are not limited to, ZANAMIVIR $^{TM}$  (2,4-dideoxy-2,3-didehydro-4-guanidino-sialic acid) and OSELTA-MIVIR $^{TM}$  (ethyl 4-acetamido-5-amino-3-(1-ethylpropxy)-1-

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cyclohexene-1-carboxylate) and known, biologically active derivatives of any of the above.

While embodiments of the invention have been illustrated and described, it is not intended that these embodiments illustrate and describe all possible forms of the invention. 5 Rather, the words used in the specification are words of description rather than limitation, and it is understood that various changes may be made without departing from the spirit and scope of the invention.

What is claimed is:

- 1. An assay complex comprising at least one test ligand and at least one composition comprising a solid support and a conjugate comprising a detectable label covalently attached to the substrate of a neuraminidase wherein modification of said substrate causes a detectable change in said label, wherein the test ligand is attached by covalent interaction with said solid support.
- 2. The complex of claim 1, wherein said detectable label is selected from the group consisting of a fluorogenic label and a chromogenic label.
- 3. The complex of claim 2, wherein said fluorogenic label is selected from the group consisting of 2'-(4-methylumbel-liferyl)- $\alpha$ -D-acetyl-neuraminic acid and fluoroscein.
- 4. The complex of claim 1, wherein the substrate is sialic acid.
- **5**. The complex of claim **1** wherein the conjugate has the formula selected from the group consisting of compounds 10, 15 and 25.
- **6**. The complex of claim **1**, wherein said at least one test ligand is a library of test ligands.
- 7. The complex of claim 1, wherein said substrate is sialic acid.
- **8**. The complex of claim **1**, wherein the conjugate is attached by covalent interaction with the solid support.
- 9. The complex of claim 1, wherein the conjugate is <sup>35</sup> attached by noncovalent interaction with the solid support.
- 10. The complex of claim 1, wherein the solid support is selected from the group consisting of a bead, a slide and a chip.
- 11. The complex of claim 1, wherein the solid support is a  $^{40}$  bead.
- 12. The complex of claim 7 wherein the conjugate has the formula selected from the group consisting of compounds 10, 15 and 25.
- 13. An assay complex comprising at least one test ligand attached to a solid support and at least one detectable label-substrate conjugate comprising a detectable label covalently attached to a first substrate of an enzyme and a quencher attached to a second substrate of said enzyme wherein modi-

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fication of said substrates causes a detectable change in said label and either said detectable label covalently attached to a first substrate of said enzyme or said quencher attached to a second substrate of an enzyme is attached to said solid support.

- **14.** The complex of claim **13**, wherein said at least one test ligand is a library of test ligands.
- 15. The complex of claim 13 wherein the detectable label covalently attached to a first substrate of an enzyme has the formula selected from the group consisting of compounds 55, 59 and 60.
- **16**. The complex of claim **13** wherein the quencher attached to a second substrate of an enzyme having the formula selected from the group consisting of compounds 64, 69, and 73.
- 17. The complex of claim 13, wherein the conjugate is attached by covalent interaction with the solid support.
- **18**. The complex of claim **13**, wherein the conjugate is attached by noncovalent interaction with the solid support.
- 19. The complex of claim 13, wherein the solid support is selected from the group consisting of a bead, a slide and a chip.
- 20. The complex of claim 13, wherein the solid support is a head.
- 21. A method of identifying an enzyme inhibitor comprising:
  - a) combining in an assay mixture the complex of claim 1 and a sufficient quantity of said enzyme to react with the substrate under noninhibitory conditions; and
  - b) detecting a change in said label in said assay mixture upon combination with said enzyme; wherein the presence of an enzyme inhibitor prevents said enzyme from modifying said substrate and causing a change in said detectable label; and
  - c) identifying the test ligand which prevents said detectable change.
- 22. The method of claim 21, wherein the enzyme is neuraminidase or sialyltransferase and the natural substrate is sialic acid.
- 23. The method of claim 21, wherein said change is a fluorescent emission
- 24. The method of claim 23, wherein the detecting step is performed by monitoring the change in fluorescence emission of fluorescein attached to sialic acid.
- 25. The method of claim 23, wherein the detecting step is performed by monitoring the change in fluorescence emission of 2'-(4-methylumberllifreyl)- $\alpha$ -D-acetyl-neuraminic acid attached to sialic acid.

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